# CO<sub>2</sub> production predicted from isocitrate dehydrogenase activity and bisubstrate enzyme kinetics in the marine bacterium \*Pseudomonas nautica\*

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ABSTRACT:  $CO_2$  production in aerobic bacteria was modeled from the time-courses of the *in vitro* activity of isocitrate dehydrogenase (IDH), bacterial protein, and the concentration of the carbon source in the cultures. The model was based on the concept of bisubstrate control of the IDH reaction throughout the exponential, steady-state, and senescent phases of the cultures. In the exponential phase, the measured rates of  $CO_2$  production and the *in vitro* IDH activity were closely coupled, but in the senescent phase, they became uncoupled. The *in vitro* IDH activity remained high even after the culture's carbon source was exhausted, while the  $CO_2$  production fell to low levels. Based on the hypothesis that this uncoupling was caused by internal substrate limitation, 2 mathematical models incorporating a bisubstrate enzyme kinetics algorithm were constructed and tested. The models predicted the rate of  $CO_2$  production throughout the different phases of the cultures with an  $r^2$  greater than 0.84.

 $KEY\ WORDS:\ Respiration \cdot Metabolism \cdot Bacteria \cdot Carbon\ dioxide \cdot Ocean\ model \cdot Isocitrate\ dehydrogenase\ activity$ 

### INTRODUCTION

Impending global climate change driven by greenhouse warming has intensified the need to understand and measure the temporal and spatial variability of biogeochemical processes in the ocean (Golden 1989). The respiratory  $CO_2$  production rate  $(R_{CO_2})$  in the ocean is one of these processes that, if better known, would improve global estimates of new production and deep-ocean  $CO_2$  sequestering. Currently, measurements of  $R_{CO_2}$  can be made on individual aquatic organisms and cultures of microorganisms where rates greater than 1 µmol  $CO_2$  min<sup>-1</sup> l<sup>-1</sup> prevail, but  $R_{CO_2}$  cannot be measured in seawater from oceanic regions

The  $CO_2$  produced by aerobically growing bacteria is largely the result of 5 enzymes, isocitrate dehydrogenase (IDH),  $\alpha$ -ketoglutarate dehydrogenase (KGDH), pyruvate dehydrogenase (PDH), malic enzyme (ME), and phosphoenolpyruvate carboxykinase (PEPCK; Walsh & Koshland 1984, Holms 1986a, b). Of these, IDH is in a unique position to exert metabolic control, because it links the Krebs cycle to the respiratory electron transfer system (Bennett & Holms 1975,

and certainly cannot be measured at rates needed for mesoscale oceanographic studies (Zirino 1985, Savenkoff et al. 1995). Since the respiratory  $R_{\rm CO_2}$  is controlled by the decarboxylases associated with the Krebs cycle (Hurley et al. 1991), we intuited that these enzymes could serve as bioindices of  $R_{\rm CO_2}$  and have begun to investigate their potential in predicting  $R_{\rm CO_2}$  in marine bacteria.

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Hurley et al. 1990a, b, 1991). Furthermore, IDH is activated and inactivated by IDH kinase/phosphatase (LaPorte & Koshland 1982, Nimmo & Nimmo 1984) suggesting that IDH activity and  $R_{\rm CO_2}$  could exhibit a coupling throughout the different growth phases of the bacterial culture. Here, we characterize this coupling of IDH with  $R_{\rm CO_2}$  in batch cultures of the marine bacterium  $Pseudomonas\ nautica$ .

This study first tested the concept that  $R_{\rm CO_2}$  is a linear function of the *in vitro* activity of IDH ( $A_{\rm IDH}$ ). Finding this concept inadequate, the investigation tested the concept that  $R_{\rm CO_2}$  is a function of the *in vivo* activities of IDH ( $V_{\rm IDH}$ ) and associated decarboxylases. This part of the investigation involved mathematical modeling. A simple IDH model based on simulated  $V_{\rm IDH}$  and its empirical relationship with  $R_{\rm CO_2}$  was constructed and compared with a complex model based on the 5 decarboxylases mentioned above.

Time-series experiments with batch cultures of Pseudomonas nautica were run.  $R_{CO_2}$ ,  $A_{IDH}$ , bacterial protein, and pyruvate were measured throughout the exponential, steady state, and senescence phases of these cultures. Uncoupling between  $A_{\text{IDH}}$  and  $R_{\text{CO}_2}$  in the senescent phase became obvious. Algorithms were developed to predict  $V_{\rm IDH}$  and to calculate the total in vivo enzymatic  $CO_2$  production rate  $(V_{CO_2})$ . These algorithms were based on the ideas that senescent-phase uncoupling between  $A_{\rm IDH}$  and  $R_{\rm CO_2}$  was caused by substrate limitation of IDH, that bisubstrate enzyme kinetics could predict  $V_{\rm IDH}$  from  $A_{\rm IDH}$ , and that  $V_{\rm CO_2}$  was a linear function of  $V_{\rm IDH}$  or was the sum of the *in vivo* activity of all 5 decarboxylases. To test these ideas, time functions of the IDH substrates (isocitrate and NADP+) were modeled from pyruvate and protein time-course data. Then, from these substrate models, from the  $V_{\mathrm{IDHH}}$  model, and from the  $A_{\mathrm{IDH}}$  time course, the  $V_{\mathrm{CO}_2}$  time profile was computed. These  $V_{\rm CO_2}$  computations compared favorably with the measured  $R_{\rm CO_2}$  time profiles.

### MATERIALS AND METHODS

**Experimental design.** To investigate the degree of association between  $A_{\rm IDH}$  and  $R_{\rm CO_2}$  in different bacterial growth stages, time-profile experiments (1 or 14 d) were run on batch cultures, maintained on pyruvate at 22°C. Samples were taken for pyruvate, protein,  $A_{\rm IDH}$ , and  $R_{\rm CO_2}$  as the cultures passed through exponential growth, steady state and senescence. The results from two 1 d experiments and one 14 d experiment are presented.

Bacteria cultures. Pseudomonas nautica (Strain 617), an oil degrading bacterium isolated from the Gulf of Fos, France (Bonin 1986, Bonin et al. 1987a, b), was cultivated aerobically at 22°C in a phosphate-buffered medium (0.33 mM, pH 7.5) containing 400 mM NaCl,

10 mM MgSO<sub>4</sub>·7H<sub>2</sub>0, 10 mM CaCl<sub>2</sub>·2H<sub>2</sub>O, 10 mM KCl, 25 mM NH<sub>4</sub>Cl, 0.01 mM FeSO<sub>4</sub>·7H<sub>2</sub>O, and 20 mM pyruvate. The original culture was kindly given to us by Dr P. Bonin of the University of Aix-Marseille. Growth was monitored from absorbance at 550 nm (OD550). OD550 and protein (M) were related by the regression equation  $M = 221.8 \, \text{OD550} - 19.9 \, (\text{n} = 21)$ . This relationship held for all culture phases with an r<sup>2</sup> of 0.987 (p < 0.001).

**Respiration measurements.**  $R_{\text{CO}_2}$  (µmol CO<sub>2</sub> min<sup>-1</sup> l<sup>-1</sup>) was measured in duplicate with a Micro-Oxymax respirometer according to Varma & Palsson (1994) and Berdalet et al. (1995). The standard error was 1.3% (n = 18).

**Biochemical sampling.** Samples (5 to 10 ml) of the bacteria culture were centrifuged for 15 min at  $10\,000 \times g$  at  $4^{\circ}$ C. A sample of the supernatant fluid was immediately transferred to an acid rinsed cryovial and stored in liquid nitrogen for pyruvate analysis. Pellets of concentrated bacteria were resuspended in 2 ml of the IDH-extraction buffer [25 mM MOPS (pH 7.5) and 167 mg ml<sup>-1</sup> lysozyme] for IDH analyses, and in 2 to 4 ml 1 N NaOH (at  $20^{\circ}$ C) for protein analysis. All samples were stored in liquid nitrogen according to Ahmed et al. (1976).

**Pyruvate measurements.** Pyruvate in the supernatant fluid was measured in its acid form at 210 nm by HPLC as described by Berdalet et al. (1995). Measurements were made in duplicate; the standard error was 1.2% (n = 10).

In vitro IDH activity ( $A_{\text{IDH}}$ ).  $A_{\text{IDH}}$  was assayed photometrically by following the NADPH production at 340 nm (Reeves et al. 1969, 1972, Holms & Bennett 1971, Berdalet et al. 1995). The assay is based on the reaction, ISO + NADP\*  $\leftrightarrow \alpha$ -KG + NADPH + CO<sub>2</sub>, where ISO and  $\alpha$ -KG represent isocitrate and  $\alpha$ -ketoglutarate. Results are reported as  $\mu$ mol CO<sub>2</sub> (min)<sup>-1</sup> (liter of culture)<sup>-1</sup>. Each data point represents 4 analyses (see Figs. 1 & 2). The standard error was 1.1% (n = 21).

**Protein measurements.** Protein in the bacterial pellet was analyzed by the Lowry method (Lowry et al. 1951) and reported as mg protein (liter of culture)<sup>-1</sup>. Bovine Serum Albumen (BSA) was used as a standard. Each data point represents 4 analyses (see Figs. 1 & 2). The standard error was 0.7 % (n = 21).

Modeling computation. Modeling was done on a Macintosh IISi computer using the software Data Desk®, version 4.2, from Data Description Inc., Ithaca, NY. The merit of the isocitrate and NADP+ algorithms was judged on their ability to generate declining time-profiles of isocitrate and NADP+ within the boundary conditions of 1 to 60  $\mu$ M (White et al. 1964, Lehninger 1970, Lehninger et al. 1993). The merit of the  $V_{\rm IDH}$  algorithm was judged on the correlation (r²) between the  $V_{\rm IDH}$  and  $R_{\rm CO_2}$  time profiles.

### RESULTS

# Time profiles of $R_{\mathrm{CO_2}}$ , $A_{\mathrm{IDH}}$ , protein, and pyruvate

The time-profile observations in the bacteria cultures during Expts A, B, and C are shown in Figs. 1 & 2. The pyruvate in the culture medium in the 3 experiments decreased from an initial level of 20 mmol l<sup>-1</sup> to nearly zero within 24 h. Protein increased as an inverse image of the pyruvate decline from about 10 mg l<sup>-1</sup> at the beginning to about 280 mg l<sup>-1</sup> at the end of the experiments (Fig. 1).

The time course of the pyruvate (*P*) decrease in Expt A (Fig. 1A) was described by the equation:

$$P(t) = [P_0/(1 + e^{\mu(t-\Phi)})] + P_f$$
 (1)

where  $P_0$  is the initial pyruvate level (20 µmol  $l^{-1}$ ),  $P_l$  is the final pyruvate level (0.11 µmol  $l^{-1}$ ),  $\mu$  is culture growth rate (0.47 divisions  $h^{-1}$ ), t is the time in hours, and  $\phi$  is time of pyruvate exhaustion (12.7 h). Eq. (1) predicted the observed pyruvate decrease well; the regression equation between the modeled pyruvate (P) and the measured pyruvate (X) was P = 1.00X + 0.084 ( $r^2 = 0.997$ ).

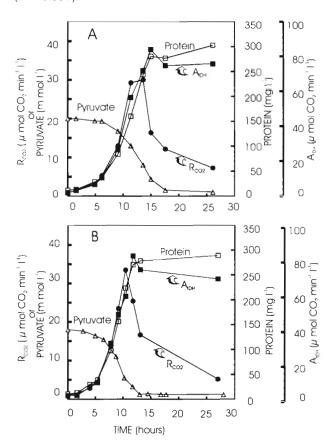


Fig. 1. Replicate 1 d time-course experiments (A & B) showing observations of  $A_{\rm 1DH}$ ,  $R_{\rm CO_2}$ , protein, and pyruvate. The analytical errors are given in the 'Materials and methods'

Table 1.  $R_{\rm CO_2}$ - $A_{\rm IDH}$  and  $R_{\rm CO_2}$ -M relationships by linear regression analysis on the exponential-phase data from Expts A & B. The  $A_{\rm IDH}$ -M regression analysis was run on the data from all 3 growth phases of Expts A & B (data are shown in Fig. 1). Slope of the regression equation (m) and its intercept (b) are given with their standard errors

m	b	n	$\mathbf{r}^2$
$0.53 \pm 0.02$	$-0.45 \pm 0.67$	11	0.984
$0.16 \pm 0.08$	$-0.50 \pm 0.84$	11	0.978
$0.28 \pm 0.01$	$0.86 \pm 1.90$	21	0.974
	$0.53 \pm 0.02$ $0.16 \pm 0.08$	$0.53 \pm 0.02$ $-0.45 \pm 0.67$ $0.16 \pm 0.08$ $-0.50 \pm 0.84$	$0.53 \pm 0.02$ $-0.45 \pm 0.67$ 11 $0.16 \pm 0.08$ $-0.50 \pm 0.84$ 11

The time course of the protein (M) increase in Expt A (Fig. 1A) was described by :

$$M(P) = M_0 + \gamma (P_\alpha - P) \tag{2}$$

where M is the bacterial protein,  $M_0$  is the observed protein immediately after inoculation (11.6 mg protein  $l^{-1}$  for Expt A),  $\gamma$  is the yield of bacterial protein when all the pyruvate has been consumed [15.7 mg protein (mmol pyruvate)<sup>-1</sup> for Expt A], P is the pyruvate level from Eq. (1), and  $P_\alpha$  is P at t=0 (from Eq. 1). Note that since P=f(t) and M=f(P), Eq. (2) is also a time function. The regression equation between the modeled protein (M) and the measured protein (X) was M=1.00X-12.58 ( $r^2=0.965$ ).

The time course of the  $A_{\rm JDH}$  increase in both experiments followed the protein time course throughout all phases of growth. The regression between the 2 variables is given in Table 1.

The  $R_{\rm CO_2}$  time profiles did not follow the  $A_{\rm IDH}$  and protein time profiles over the life span of the cultures (Table 1, Figs. 1 & 2). In exponential growth,  $A_{\rm IDH}$  and  $R_{\rm CO_2}$  were in parallel and highly correlated (Table 1), but in senescence during pyruvate limitation,  $A_{\rm IDH}$  and  $R_{\rm CO_2}$  diverged. This divergence was especially apparent in Expt C where even after 13 d of nutrient depletion,  $A_{\rm IDH}$  maintained at least 50% of its maximum activity whereas  $R_{\rm CO_2}$  had fallen to less than 5% of its maximum value (Fig. 2).

# Respiration models

The experimental results (Figs. 1 & 2) indicate clearly that  $R_{\rm CO_2}$  cannot be predicted in a simple linear fashion by changes in  $A_{\rm IDH}$  alone. If changing IDH concentrations ( $\approx A_{\rm IDH}$ ) were the important regulatory mechanism for  $R_{\rm CO_2}$  in *Pseudomonas nautica*, then  $A_{\rm IDH}$  would have declined along with  $R_{\rm CO_2}$  in senescence. Since  $A_{\rm IDH}$  did not decline, an alternative hypothesis was needed to explain the observations. If  $R_{\rm CO_2}$  were regulated through changes in the *in vivo* IDH activity ( $V_{\rm IDH}$ ), then a decline in the intracellular isocitrate and

Table 2. Parameters and constants used in the pyruvate and protein Eqs. (1 & 2) and in the simple IDH and 5 enzyme models (Table 3) to predict  $V_{CO_2}$  (Figs. 4 to 6) from observations in Expt A (Fig. 1A)

Paramete	er Value	Units
a	$1.3 \times 10^{-2}$	(mmol pyruväte l <sup>-1</sup> ) <sup>-1</sup>
b	$4.3 \times 10^{-5}$	$(mg protein l^{-1})^{-1}$
χ	0.19	
δ	$8.76 \times 10^{-4}$	µmol NADP+ (mmol pyruvate)-1
φ	12.72	h
γ	15.7	mg protein (mmol pyruvate)-1
η	$5.9 \times 10^{-7}$	µmol isocitrate (mg protein)-1
$K_{\text{ld}}$	42.8	μmol
$K_{\mathrm{ISO}}$	8.75	μmol isocitrate
$K_{NADP^*}$	18.0	μmol NADP <sup>+</sup>
λ	$4.44 \times 10^{-4}$	µmol isocitrate (mmol pyruvate)
μ	0.471	divisions h 1
$M_{\rm o}$	11.7	mg protein l <sup>-1</sup>
ν	1.2	
$P_{\sigma}$	20.06	mmol pyruvate l <sup>-1</sup>
$P_{\mathfrak{f}}$	0.11	mmol pyruvate l <sup>-1</sup>
$P_0$	20.00	mmol pyruvate l <sup>-1</sup>
Q	$8.17 \times 10^{-7}$	mmol pyruvate (mg protein) <sup>-2</sup> l <sup>-1</sup>
θ	$2.05 \times 10^{-5}$	$(mg protein l^{-1})^{-1}$
σ	$7.6 \times 10^{-3}$	h-1
υ	1.51	mmol pyruvate l <sup>-1</sup>
ω	$5.2 \times 10^{-5}$	μmol NADP+ (mg protein)-1
Ψ	$7.32 \times 10^{-3}$	(mg protein l <sup>-1</sup> ) <sup>-1</sup>
ζ	$8.32 \times 10^{-2}$	(mmol pyruvate l <sup>-1</sup> ) <sup>-1</sup>

NADP+ pools falling in parallel with the declining extracellular pyruvate (Fig. 1) would reduce  $V_{\rm IDH}$  in senescence even though  $A_{\rm IDH}$  remained high. Such a hypothesis would predict a lower  $V_{\rm IDH}$  to  $A_{\rm IDH}$  ratio, or a lower value of its physiological expression,  $R_{\rm CO_2}/A_{\rm IDH}$ , in senescence than in exponential growth. The observations support this hypothesis.  $R_{\rm CO_2}/A_{\rm IDH}$  ranged from 0.095 to 0.006 during senescence (Fig. 1) while during exponential growth it was 0.53  $\pm$  0.02 (Table 1).

Stronger support for this hypothesis would be a demonstration of how  $A_{\rm IBH}$  could be modulated by enzyme kinetics to give a  $V_{\rm IDH}$  time course that moved in parallel with the measured  $R_{\rm CO_2}$  time course. Here we demonstrate how this can be done by developing 2 respiration models based on the data of Expt A.

To simulate the  $V_{\rm IDH}$  time course from the  ${\rm A}_{\rm IDH}$  measurements in Expt A one needs to consider bisubstrate enzyme kinetics because 2 substrates (NADP+ and isocitrate) are the required reactants in the IDH catalyzed reaction (Hurley et al. 1991, Kuby 1991). The following equation in which  $V_{\rm IDH} = f(A_{\rm IDH})$  can be written to describe the IDH reaction:

$$V_{\text{IDH}} = \frac{vA_{\text{IDH}}}{1 \cdot \left(\frac{K_{\text{ISO}}}{|\text{ISO}|} + \left(\frac{K_{\text{NADP}^{*}}}{|\text{NADP}^{*}|}\right) + \left((K_{\text{IG}}K_{\text{NADP}^{*}})/|\text{ISO}|[\text{NADP}^{*}]\right)}$$

Here,  $K_{\rm ISO}$  and  $K_{\rm NADP^+}$  are bisubstrate-Michaelis constants for isocitrate and NADP+ (Mahler & Cordes 1971, Kuby 1991),  $K_{\rm ia}$  is the dissociation constant for the enzyme-isocitrate complex, and [ISO] and [NADP+] are the substrate concentrations (Table 2). In the 2 models developed in this paper the enzyme activities ( $V_{\rm IDH}$  and  $A_{\rm IDH}$ ) are normalized by culture volume and both are functions of time (Fig. 1). Thus  $A_{\rm IDH}$  in Eq. (3) is not a constant; it was measured as a time function (Fig. 1), but it also equals  $MV_{\rm max}$ , where  $V_{\rm max}$  is the maximal velocity of IDH and M is the protein from Eq. (2). v is a unitless scaling factor that reflects an enzyme assay's underestimation of  $V_{\rm max}$ : Walsh & Koshland's (1984) value of 1.2 is used here for v.

From the conceptual model that  $R_{\rm CO_2} = f(V_{\rm IDH})$  and that  $V_{\rm IDH}$  can be calculated from Eq. (3), 2 models of CO<sub>2</sub> production were developed. The first model was based on the isocitrate and NADP<sup>+</sup> time profiles (Fig. 3) which were modeled from Eqs. (1 & 2) for the pyruvate and protein time profiles and from the empirical relationship between  $R_{\rm CO_2}$  and  $V_{\rm IDH}$  throughout the time

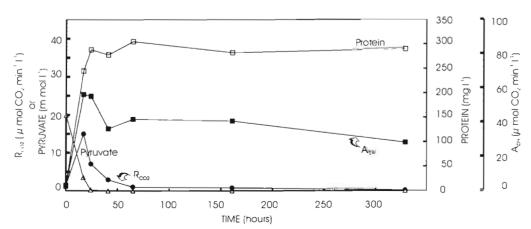


Fig. 2. 14 d time-course of  $A_{\text{IDH}_{1}}$ ,  $R_{\text{CO}_{2}}$ , protein, and pyruvate (Expt C)

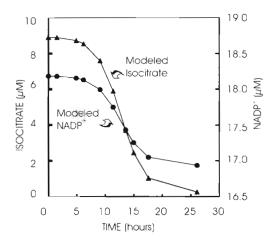


Fig. 3. Simulated isocitrate and NADP\* concentrations from Eqs. (4 & 5) for Expt A

course of Expt A (Fig. 4B). The second, more complex model computes  $V_{\rm IDH}$  as in the first model, but replaces the empirical relationship between  $R_{\rm CO_2}$  and  $V_{\rm IDH}$  by a simulation of the contribution to  $R_{\rm CO_2}$  by the *in vivo* activity of the 4 other decarboxylases, namely,  $V_{\rm KGDH}$ ,  $V_{\rm PDH}$ ,  $V_{\rm ME}$ , and  $V_{\rm PEPCK}$ .

### IDH model

If the low  $R_{\rm CO_2}$  in senescence is caused by substrate limitation, then Eq. (3) should predict the shape of the  $R_{\rm CO_2}$  versus time curve throughout the exponential, steady state, and senescent phases of the bacteria culture. To test this we need values of  $K_{\rm ISO}$ ,  $K_{\rm NADP^+}$ ,  $K_{\rm Ia}$  and the time courses of isocitrate and NADP<sup>+</sup>. The Michaelis constants,  $K_{\rm ISO}$  and  $K_{\rm NADP^+}$ , for bacteria range from 0.4 to 250  $\mu$ M for isocitrate, and 0.3 to 27  $\mu$ M for NADP<sup>+</sup> (Walsh & Koshland 1984, Chen & Gadal 1990). For our model we set  $K_{\rm ISO}$  equal to

8.75  $\mu$ M and  $K_{NADW^+}$  equal to 18.0  $\mu$ M (Table 2). These values are similar to those used by Hurley et al. (1990a, b, 1991).  $K_{Ia}$  has not been measured in bacteria, so we set it to 42.8  $\mu$ M through optimization. The concentrations of isocitrate and NADP+ were made functions of both pyruvate in the growth media (P) and the bacterial protein (M). Isocitrate concentrations between the boundary conditions, 8.9 and 0.62  $\mu$ M (Fig. 3), are predicted from the equation:

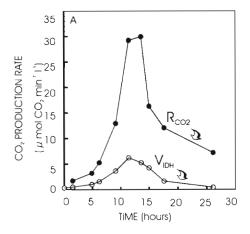
$$ISO(P, M) = \lambda P + \eta M \tag{4}$$

where  $\lambda$  is  $4.44 \times 10^{-4}$  µmol isocitrate (mmol pyruvate)<sup>-1</sup> and  $\eta$  is  $5.9 \times 10^{-7}$  µmol isocitrate (mg protein)<sup>-1</sup>. NADP+ concentrations between the boundary conditions, 18.13 and 16.58 µM (Fig. 3), are predicted from the equation:

$$NADP^{+}(P, M) = \delta P + \omega M \tag{5}$$

where  $\delta$  is  $8.76 \times 10^{-4}$  µmol NADP+ (mmol pyruvate)<sup>-1</sup> and  $\omega$  is  $5.2 \times 10^{-5}$  µmol NADP+ (mg protein)<sup>-1</sup>. Fig. 3 shows the time courses of these functions for Expt A. Note that they both resemble the pyruvate time course (Fig. 1A). The algorithms for pyruvate and protein (Eqs. 1 & 2) were used in Eqs. (4 & 5) to calculate the isocitrate and NADP+ time functions (Fig. 3). Because Eqs. (1 & 2) represent the original protein and pyruvate measurements so well, using either Eqs. (1 & 2) or using the original data (Fig. 1) changes the calculation of isocitrate and NADP+ very little.

When the isocitrate and NADP+ functions (Eqs. 4 & 5) and the values of  $K_{\rm ISO}$ ,  $K_{\rm NADP+}$ , and  $K_{\rm ia}$  (given above) are used in Eq. (3),  $V_{\rm IDH}$  ranges from 0.3  $\mu$ M min<sup>-1</sup> l<sup>-1</sup> at the beginning of Expt A to a maximum of 6.1  $\mu$ M min<sup>-1</sup> l<sup>-1</sup> at 11.4 h (Fig. 4A). By the end of the experiment  $V_{\rm IDH}$  has dropped to 0.9 mM min<sup>-1</sup> l<sup>-1</sup> From Fig. 4A one can see that throughout all phases of the bacteria culture, the  $V_{\rm IDH}$  time course resembles the  $R_{\rm CO_2}$  time course. A plot of  $V_{\rm IDH}$  against  $R_{\rm CO_2}$  (see Fig. 4B) in Expt A, shows a strong correlation (r<sup>2</sup> = 0.877) with a



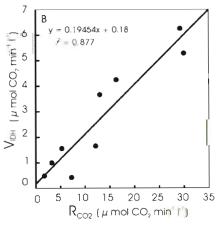
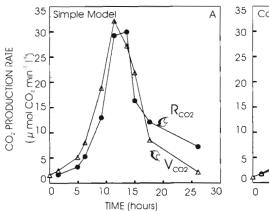


Fig. 4. (A)  $V_{\rm IDH}$  generated from Eq. (3) counterpoised against the  $R_{\rm CO_2}$  from Expt A. (B)  $V_{\rm IDH}$  regressed against  $R_{\rm CO_2}$ ; slope is  $\chi$ . All calculations based on data from Expt A



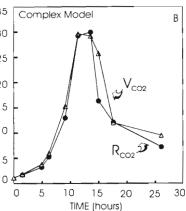


Fig. 5. For Expt A: correspondence between  $CO_2$  production rate predicted by the models  $(V_{CO_2})$  and measured  $CO_2$  production rate  $(R_{CO_2})$ . Parameters used in the models are summarized in Table 2, equations are summarized in Table 3

 $V_{\rm IDH}$  to  $R_{\rm CO_2}$  ratio (slope) of 0.19. This leads to a calculation of  $V_{\rm CO_2} = f(V_{\rm IDH})$  from the final algorithm of the simple IDH model, i.e.:

$$V_{\rm CO_2} = V_{\rm IDH}/\chi$$
 (6)

where  $\chi = V_{\rm IDH}/R_{\rm CO_2}$  and  $V_{\rm CO_2}$  is the predicted rate of CO<sub>2</sub> production (as contrasted to  $R_{\rm CO_2}$ ). Fig. 5A shows the resemblance of  $V_{\rm CO_2}$  to  $R_{\rm CO_2}$ . The regression equation between  $V_{\rm CO_2}$  and  $R_{\rm CO_2}$ , for Expt A, is  $V_{\rm CO_2} = 1.001\,R_{\rm CO_2} + 0.93$  (n = 9, r² = 0.877).

### Five enzyme model

The simple IDH model, in spite of its ability to simulate the  $R_{\rm CO_2}$  time course, simplifies the role of KGDH, PDH, ME, PEPCK to a constant,  $\chi$ . A more realistic model would simulate the *in vivo* activity of these enzymes. To achieve this we developed a more complex model based on the equation:

$$V_{\text{CO}_2} = V_{\text{IDH}} + V_{\text{KGDH}} + V_{\text{PDH}} + V_{\text{ME}} + V_{\text{PEPCK}}$$
 (7)

where V represents the  $in\ vivo$  activity of each decarboxylase in  $\mu mol\ min^{-1}$  (liter of culture)<sup>-1</sup> (Walsh & Koshland 1984). Algorithms for  $V_{\rm KGDH}$ ,  $V_{\rm PDH}$ ,  $V_{\rm ME}$ , and  $V_{\rm PEPCK}$  as functions of  $V_{\rm IDH}$  are developed as follows.

 $V_{\rm KGDH}$  is closely coupled to  $V_{\rm IDH}$ . For *Escherichia coli* growing on glucose-6-phosphate,  $V_{\rm KGDH}$  is 73% of  $V_{\rm IDH}$ ; for *E. coli* growing on acetate,  $V_{\rm KGDH}$  is 94 to 95% of  $V_{\rm IDH}$ . We developed a  $V_{\rm KGDH}$  algorithm assuming that  $V_{\rm KGDH}$  in *Pseudomonas nautica* behaves as it does in *E. coli*. The algorithm balances the effect of the pyruvate decrease against the protein increase so that the  $V_{\rm KGDH}$  lags behind the  $V_{\rm IDH}$  during growth on pyruvate but becomes equal to it when pyruvate exhaustion limits growth. Initially,  $V_{\rm KGDH}/V_{\rm IDH}$  is 0.74, at the maximum it is 0.84, and at the end of the experiment it is 1.0 (Fig. 6). This algorithm takes the form:

$$V_{\text{KGDH}}(V_{\text{IDH}}, P, M) = V_{\text{IDH}}(1 - aP - bM) \tag{8}$$

where the constants, a and b, equal  $1.3 \times 10^{-2}$  (mmol pyruvate  $l^{-1}$ )<sup>-1</sup> and  $4.3 \times 10^{-5}$  (mg protein  $l^{-1}$ )<sup>-1</sup>, respectively.

The PDH algorithm (Eq. 9) is dictated by the requirement that  $V_{\rm PDH}$  exceed  $V_{\rm IDH}$  when the bacteria use glucose or pyruvate as a carbon source, because some carbon departs the Krebs cycle at the level of citrate to supply citrate lyase for fatty acid synthesis. Well-oxygenated glucose-grown cultures of *Escherichia coli* are characterized by a  $V_{\rm PDH}/V_{\rm IDH}$  ratio of 1.66 (Holms 1986a, b). After a culture exhausts its pyruvate and grows on acetate, the  $V_{\rm PDH}/V_{\rm IDH}$  ratio would be lower

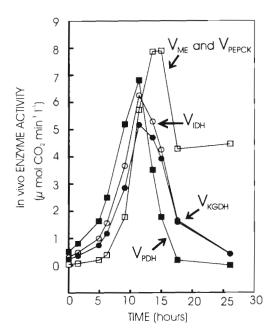


Fig. 6. In vivo activities of IDH, KGDH, PDH, ME, and PEPCK as calculated from Eqs. (3), (8), (9), and (10). The  $A_{\rm IDH}$ , pyruvate, and protein data from which these curves were calculated are shown in Fig. 1A (Expt A). The modeled isocitrate and NADP\* inputs for Eq. (3) are shown in Fig. 3. The sum of these activities from Eq. (7) ( $V_{\rm CO_2}$ ) is shown in Fig. 5B

than 1. We designed a function, based on the time courses of  $V_{\rm IDH}$ , pyruvate, and protein in Expt A, that would fall off from the maximum faster than the  $V_{\rm IDH}$  function (Fig. 6), but during exponential growth, while the medium still contains pyruvate,  $V_{\rm PDH}$  would range from 1.66 to  $1.28 \times V_{\rm IDH}$ . In senescence,  $V_{\rm PDH}$  would be smaller than  $V_{\rm IDH}$  (Fig. 6). This algorithm takes the form:

$$V_{\text{PDH}}(V_{\text{IDH}}, P, M) = V_{\text{IDH}}(\zeta P - \theta M - QM^2/P)$$
 (9)

in which the constants are:  $\zeta=8.32\times10^{-2}$  (mmol pyruvate  $l^{-1})^{-1}$ ,  $\theta=2.05\times10^{-5}$  (mg protein  $l^{-1})^{-1}$ , and  $Q=8.17\times10^{-7}$  (mmol pyruvate) (mg protein) $^{-2}$   $l^{-1}$ .

 $V_{\rm PEPCK}$  was set equal to  $V_{\rm ME}$  as in the model of Walsh & Koshland (1984). The algorithm for ME (Eq. 10) was designed to increase  $V_{\rm ME}$  after pyruvate depletion when the cells utilize excreted acetate (Holms 1986a, b). The algorithms for both enzymes generate low activities while pyruvate levels are high. This is consistent with the pyruvate inhibition of ME observed by Hsu et al. (1967). An important characteristic of these enzymes is that they retain  $in\ vivo$  activity in the senescent phase (Holms 1986a, b; Fig. 6). The resulting time course of the algorithm is shown in Fig. 6. The algorithm itself takes the form:

$$V_{\rm ME}(V_{\rm 1DH},P,M)=V_{\rm PEPCK}=V_{\rm IDH}e^{-\sigma t}(\psi M+\upsilon/P)$$
 (10) where  $\psi=7.32\times10^{-3}$  l (mg protein)<sup>-1</sup>,  $\upsilon=1.51$  mmol pyruvate l<sup>-1</sup> and  $\sigma=7.6\times10^{-3}$  h<sup>-1</sup>.

Eq. (7), summing the *in vivo* activities of the 5 enzymes (Fig. 6), computes  $V_{\rm CO_2}$ . This sum should equal  $R_{\rm CO_2}$ . For Expt A, the time course for  $V_{\rm CO_2}$  is shown in Fig. 5B. It simulates closely the  $R_{\rm CO_2}$  time pro-

file from Fig. 1A. When  $V_{\rm CO_2}$  is plotted against  $R_{\rm CO_2}$ , a 1:1 relation results ( $V_{\rm CO_2}=0.998R_{\rm CO_2}+1.69$ , n = 9) with  $\rm r^2=0.920$ . This complex 5 enzyme model (Fig. 5B) predicts both the shape and the magnitude of the  $R_{\rm CO_2}$  time course as well as the simple IDH model (Fig. 5A), but it does not invoke the empirical relationship between  $V_{\rm IDH}$  and  $R_{\rm CO_2}$  ( $\chi$ ). The equations for both models are summarized in Table 3.

### DISCUSSION

# Isocitrate dehydrogenase activity

The behavior of  $A_{\rm IDH}$  in Pseudomonas nautica throughout different bacterial growth phases was unknown prior to this investigation. Would the absence of a carbon source signal the cell to dismantle surplus enzyme systems? Would the cell cannibalize its Krebs cycle enzymes and use the materials for other purposes, if there were no organic molecules to oxidize? Would IDH be inactivated by IDH kinase/phosphatase and kept in reserve, or would it be kept active but substrate limited? In reviewing recent research on the general question of maintaining enzymatic capacity during adverse conditions, Nystrom et al. (1990) found that most marine Vibrio strains remain metabolically active during starvation and are primed for substrate availability. For marine bacteria that inhabit an environment characterized by low but highly variable bioavailability of organic nutrients, this ability to maintain a reserve of alert-status enzymes would be important.

Table 3. Summary of equations used in both models. Each equation is a time function because it depends directly or indirectly upon the pyruvate time function (Eq. 1) and the  $A_{\rm IDH}$  time course (Fig. 1). Note that  $V_{\rm PEPCK} = V_{\rm ME}$ . For the simple IDH model the first 6 equations are used. For the 5 enzyme model, all equations except Eq. (6) are used. Table 2 lists parameter values. Variable and parameters are defined in the text

Equation	Source in this paper:		
	Eq. no.	Fig. no.	
$P(t) = [P_0/(1 + e^{\mu(t - \Phi)})] + P_1$	(1)	1	
$M(P) = M_0 + \gamma(P_\alpha - P)$	(2)	1	
$V_{\text{IDH}} = \frac{VA_{\text{IDH}}}{1 + \left(\frac{K_{\text{ISO}}}{ \text{ISO} }\right) + \left(\frac{K_{\text{NADP}^*}}{ \text{NADP}^* }\right) + \left((K_{\text{ia}}K_{\text{NADP}^*})/[\text{ISO}][\text{NADP}^*]\right)}$	(3)	4	
$SO(P, M) = \lambda P + \eta M$	(4)	3	
$NADP^{+}(P, M) = \delta P + \omega M$	(5)	3	
$V_{\rm CO_2} = V_{\rm IDH}/\chi$	(6)	5	
$V_{\text{COp}} = V_{\text{IDH}} + V_{\text{KGDH}} + V_{\text{PDH}} + V_{\text{ME}} + V_{\text{PEPCK}}$	(7)	5	
$V_{\text{NGDH}}(V_{\text{IDH}}, P, M) = V_{\text{IDH}} (1 - aP - bM)$	(8)	6	
$V_{\text{I-DPH}}(V_{\text{IDH}}, P, M) = V_{\text{IDH}}(\zeta P - \theta M - QM^2/P)$	(9)	6	
$V_{\text{NMF}}(V_{\text{IDH}}, P, M) = V_{\text{PERICK}} = V_{\text{IDH}} e^{-\sigma t} (\psi M + \upsilon / P)$	(10)	6	

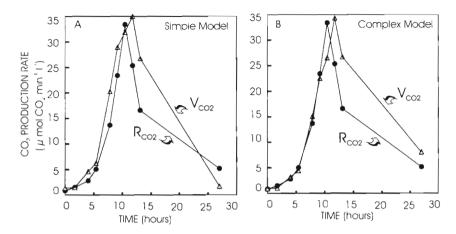


Fig. 7. Model simulations for Expt B (Fig. 1B). The correspondence between  $V_{\rm CO_2}$  predicted by the (A) simple IDH and (B) complex 5 enzyme models and  $R_{\rm CO_2}$  is given by the regression equation  $V_{\rm CO_2} = 1.19R_{\rm CO_2} + 0.096$  ( $\rm r^2 = 0.89$ ) for the simple model and  $V_{\rm CO_2} = 0.99R_{\rm CO_2} + 1.67$  ( $\rm r^2 = 0.84$ ) for the complex model

In our experiments,  $A_{\rm IDH}$  tracked collular protein throughout the exponential and steady states, and, to a lesser extent, the senescent phases (Table 1, Fig. 1). This suggests that IDH is constitutive in *Pseudomonas nautica*, that during starvation this enzyme is in alert status, and that it is not inactivated or degraded by IDH kinase/phosphatase after the culture's carbon source has been exhausted. It also suggests that  $A_{\rm IDH}$  could serve as an index of living biomass in the absence of a protein measurement. When the  $A_{\rm IDH}$  and protein data from both Expts A and B were pooled (Table 1) the regression equation was:  $A_{\rm IDH} = 0.28M + 0.86$  ( $\rm r^2 = 0.975$ ).

No simple linear equation could relate  $A_{\rm IDH}$  to  $R_{\rm CO_2}$  throughout all phases of the bacteria culture;  $A_{\rm IDH}$  correlated well with  $R_{\rm CO_2}$  only in the exponential growth phase (Fig. 1). The magnitude of  $A_{\rm IDH}$  exceeded  $R_{\rm CO_2}$  at all times, demonstrating the great capacity for  ${\rm CO_2}$  production at the enzyme level.

# Reconstruction of the respiration rate

The reactions responsible for  $R_{\rm CO_2}$  have long been identified, but the control of these reactions and their kinetics are still not well enough understood to allow quantitative reconstruction of the physiological respiration rate ( $R_{\rm CO_2}$ ). Rare is the account in the literature where either  $R_{\rm CO_2}$  or  $R_{\rm O_2}$  of whole cells or cultures is calculated from first principles and measurements of enzyme activities, substrate concentrations and biomass levels. In this paper we demonstrate how this might be done using Eqs. (4) to (9). We show that from the  $A_{\rm IDH}$ , pyruvate and protein time-course data (Fig. 1A), one can simulate the physiological respiration rate ( $R_{\rm CO_2}$ ). The degree of success was high for Expt A (Fig. 1); the  $r^2$  values for the simple IDH and complex 5 enzyme models were 0.88 and 0.92.(Fig. 5)

Similar success was duplicated for an independent experiment, presented in Fig. 1B, Expt B. For this

experiment, the measured values of  $P_0$ ,  $M_0$ , and  $\gamma$  were 18 mmol pyruvate  $l^{-1}$ , 10 mg protein  $l^{-1}$ , and 16.2 mg protein (mmol pyruvate) $^{-1}$ . When these values and the  $A_{\rm IDH}$  data from Fig. 1B were entered into the 2 models (Table 3) and all other parameters held at the values listed in Table 2,  $R_{\rm CO_2}$  was predicted with an  $r^2$  value of 0.89 for the simple model and 0.84 for the complex model (Fig. 7). This success demonstrates the feasibility of using bisubstrate enzyme kinetics to predict the physiological respiration rate of a bacteria culture from 3 variables defining the culture's physiological state, namely, the enzyme potential ( $A_{\rm IDH}$ ), the biomass (M), and the level of its carbon source (P).

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# LITERATURE CITED

Ahmed SI, Kenner RA, King FD (1976) Preservation of enzyme activity in marine plankton by low-temperature freezing. Mar Chem 4:133–139

Bennett PM, Holms WH (1975) Reversible inactivation of the isocitrate dehydrogenase of *Escherichia coli* ML 308 during growth on acetate. J Gen Microbiol 87:37–51

Berdalet E, Packard T, Lagacé B, Roy S, St-Amand L, Gagné JP (1995) CO<sub>2</sub> production, O<sub>2</sub> consumption and isocitrate dehydrogenase in the marine bacterium *Vibrio natriegens*. Aquat Microb Ecol 9:211–217

Bonin P (1986) Nitrate reduction in simulated microniches by a denitrifying marine bacterium. Can J Microbiol 33: 276-279

Bonin P, Bertrand JC, Giordano G, Gilewicz M (1987a) Specific sodium dependence of a nitrate reductase in a marine bacterium. FEMS Microbiol Lett 48:5–9

Bonin P, Gilewicz M, Bertrand JC (1987b) Denitrification by a marine bacterium *Pseudomonas nautica* strain 617 Ann Inst Pasteur/Microbiol 138:371~383

Chen RD, Gadal P (1990) Structure, function and regulation of NAD and NADP-dependent isocitrate dehydrogenase in

- higher plants and other organisms. Plant Physiol Biochem 28:411–427
- Golden F (1989) The role of the seas in the planetary hothouse. Oceanus 32:2-3
- Holms WH (1986a) The central metabolic pathways of Escherichia coli: relationship between flux and control at a branch point, efficiency of conversion to biomass, and excretion of acetate. Curr Top Cell Regul 28:69–105
- Holms WH (1986b) Evolution of the glyoxylate bypass in Escherichia coli—an hypothesis which suggests an alternative to the Krebs cycle. FEMS Microbiol Lett 34: 123-127
- Holms WH, Bennett PM (1971) Regulation of isocitrate dehydrogenase activity in *Escherichia coli* on adaptation to acetate. J Gen Microbiol 65:57–68
- Hsu RY, Lardy HA, Cleland WW (1967) Pigeon liver malic enzyme: V. Kinetic studies. J Biol Chem 242:5315–5322
- Hurley JH, Dean AM, Koshland DE Jr, Stround RM (1991) Catalytic mechanism of NADP-dependent isocitrate dehydrogenase, implications from the structures of magnesium isocitrate and NADP complexes. Biochemistry 30: 8671-8678
- Hurley JH, Dean AM, Sohl JL, Koshland DE Jr, Stroud RM (1990a) Regulation of an enzyme by phosphorylation at the active site. Science 249:1012-1016
- Hurley JH, Thorsness PE, Koshland DE Jr, Stroud RM (1990b) Regulation of isocitrate dehydrogenase by phosphorylation involves no long-range conformational change in the free enzyme. J Biol Chem 265:3599–3602
- Kuby SA (1991) Enzyme catalysis, kinetics, and substrate binding. CRC Press, Boca Raton
- LaPorte DC, Koshland DE Jr (1982) A protein with kinase and phosphatase activities involved in regulation of tricarboxylic acid cycle. Nature 300:458–460
- Lehninger AL (1970) Biochemistry. The molecular basis of cell structure and function. Worth Publishers, New York
- Lehninger AL, Nelson DL, Cox MM (1993) Principles of bio-

Responsible Subject Editor: J. T. Hollibaugh, Tiburon, California, USA

- chemistry. Worth Publishers, New York
- Lowry OH, Rosebrough NJ, Farr AL, Randall RJ (1951) Protein measurement with the Folin phenol reagent. J Biol Chem 193:265–275
- Mahler HR, Cordes EH (1971) Biological chemistry. Harper and Row, New York
- Nimmo GA, Nimmo HG (1984) The regulatory properties of isocitrate dehydrogenase kinase and isocitrate dehydrogenase phosphatase from *Escherichia coli* ML308 and the roles of these activities in the control of isocitrate dehydrogenase. Eur J Biochem 141:409–414
- Nystrom T, Flardh K, Kjelleberg S (1990) Responses to multiple-nutrient starvation in marine *Vibrio* sp. strain CCUG 15956. Bacteriol 172:7085–7097
- Reeves HC, Daumy GO, Lin CC, Houston M (1972) NADP\* specific isocitrate dehydrogenase of *Escerichia coli* 1. Purification and characterization. Biochim Biophys Acta 258:27–39
- Reeves HC, Rabin R, Wegener WS, Ajl SJ (1969) Assays of enzymes of the tricarboxylic acid and glyoxylate cycles. In:
  Norris JR (ed) Methods in microbiology. Academic Press,
  San Francisco
- Savenkoff C, Vézina AF, Chanut JP, Gratton Y (1995) Respiratory activity and CO<sub>2</sub> production rates of microorganisms in the lower St. Lawrence Estuary. Cont Shelf Res 15: 613-631
- Varma A, Palsson BO (1994) Stoichiometric flux balance models quantitatively predict growth and metabolic by-product secretion in wild-type *Escherichia coli* W3110. Appl Environ Microbiol 60:3724–3731
- Walsh K, Koshland DE Jr (1984) Determination of flux through the branch point of two metabolic cycles. J Biol Chem 259:9646-9654
- White A, Handler P, Smith EL (1964) Principles of biochemistry. McGraw-Hill, New York
- Zirino A (1985) Mapping strategies in chemical oceanography. American Chemical Society, Washington, DC

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