

# Vapor-liquid equilibria of binary mixtures of methyl alkanoates + isomeric butanols

Ortega, J.; Susial, P.

Laboratorio de Termodinamica y Fisicoquimica, Escuela Superior de Ingenieros Industriales, Universidad de Las Palmas de Gran Canaria, 35071 - Las Palmas de Gran Canaria, Canary Islands, Spain

(Received in final form July 20, 1994)

*Isobaric vapor-liquid measurements are reported at 101.32 10<sup>3</sup>Pa for methyl ethanoate, methyl propanoate, and methyl butanoate + 1-butanol, + 2-butanol, or + 2-methyl-1-propanol.*

## 1. INTRODUCTION

In continuation of the experimental studies carried out in our laboratory on vapor-liquid equilibria (VLE) of mixtures of esters and alcohols (SUSP0890, ORTJ0905, ORTJ0906, ORTJ0907, ORTJ0934), we present here isobaric measurements at 101.32 10<sup>3</sup>Pa for methyl ethanoate, methyl propanoate, and methyl butanoate + isomeric butanols. Only a few data appear in the literature (WICI0930) for this class of systems (PATV1771, BELV0870, PERV0800, FERJ1870). Data for methyl propanoate + 2-methyl-1-propanol (SUSP0930) or + 1-butanol (SUSP0931) have been published previously. The purpose of these measurements is to compare the experimental data with those predicted by various group contribution models (see, e. g., ORTJ0934), and especially to study the effect of branching of the alkanol and of the position of the hydroxyl group.

## 2. EXPERIMENTAL SECTION

### 2.1. Apparatus and Procedure

An all-glass dynamic equilibrium still consisting of a boiling flask, Cottrell pump, and condenser was used (ORTJ0860). Equilibrium temperature  $T$  was measured by means of a high-precision Pt resistance thermometer (Comark-6800, UK) calibrated against IPTS-68 to within  $\sigma(T)/K = 0.02$ . Pressure  $P$  was measured using a quartz Bourdon tube pressure gauge (MKS Inst., Model 270 B, France) to within  $\sigma(P)/Pa = 20$ .

Samples of liquid and condensed vapor were taken at intervals of 30-40 min. The liquid-phase  $x$  and vapor-phase  $y$  mole fractions were determined with a digital densimeter (Anton Paar, Model DMA-55, Austria) to within  $\sigma(x_i) = 0.001$  and  $\sigma(y_i) = 0.002$ . The calibration curves were based on densities of synthetically prepared mixtures.

### 2.2. Materials

**C<sub>3</sub>H<sub>6</sub>O<sub>2</sub>, Methyl ethanoate** (Methyl acetate). Fluka AG (Buchs, Switzerland) "purum" grade material of stated purity > 99 mole %, degassed ultrasonically, dried over molecular sieves Type 3A (reference 69828, from Fluka), and used without further purification.  $n(D, 298.15\text{ K}) = 1.3589$ ;  $\rho_i(298.15\text{ K})/\text{kg m}^{-3} = 927.07$ .

**C<sub>4</sub>H<sub>8</sub>O<sub>2</sub>, Methyl propanoate** (Methyl propionate). Fluka AG (Buchs, Switzerland) "purum" grade material of stated purity > 99 mole %, purified as above.  $n(D, 298.15\text{ K}) = 1.3745$ ;  $\rho_i(298.15\text{ K})/\text{kg m}^{-3} = 908.86$ .

**C<sub>4</sub>H<sub>10</sub>O, 1-Butanol** (Butyl alcohol). Fluka AG (Buchs, Switzerland) "puriss p. a." grade material of stated purity > 99.5 mole % purified as above.  $n(D, 298.15\text{ K}) = 1.3974$ ;  $\rho_i(298.15\text{ K})/\text{kg m}^{-3} = 805.93$ .

**C<sub>4</sub>H<sub>10</sub>O, 2-Butanol** (sec-Butyl alcohol). Fluka AG (Buchs, Switzerland) "puriss p. a." grade material of stated purity > 99.0 mole % purified as above.  $n(D, 298.15\text{ K}) = 1.3953$ ;  $\rho_i(298.15\text{ K})/\text{kg m}^{-3} = 802.36$ .

**C<sub>4</sub>H<sub>10</sub>O, 2-Methyl-1-propanol** (Isobutyl alcohol). Fluka AG (Buchs, Switzerland) "puriss p. a." grade material of stated purity > 99.5 mole % purified as above.  $n(D, 298.15\text{ K}) = 1.3939$ ;  $\rho_i(298.15\text{ K})/\text{kg m}^{-3} = 797.83$ .

**C<sub>5</sub>H<sub>10</sub>O<sub>2</sub>, Methyl butanoate** (Methyl butyrate). Fluka AG (Buchs, Switzerland) "purum" grade material of stated purity > 99.0 mole %, purified as above.  $n(D, 298.15\text{ K}) = 1.3849$ ;  $\rho_i(298.15\text{ K})/\text{kg m}^{-3} = 892.61$ .

## 3. RESULTS

All the isobaric measurements were performed at  $(101.32 \pm 0.01)10^3\text{Pa}$ . The direct experimental values are tabulated and graphed in the Appendix and saved on disk as Standard ELDATA Files ORTJ0950.001 through ORTJ0950.018.

## REFERENCES

- BELV0870 - Belousov, V. P.; Gkhoneimi, Kh. F.; Shulgin, I. L.; Sobolev, D. M. *Zh. Prikl. Khim. (Leningrad)* 1987, 60, 1788.  
 FERJ1870 - Fernandez, J.; Berro, C.; Peneloux, A. J. *Chem. Eng. Data* 1987, 32, 17.  
 ORTJ0860 - Ortega, J.; Pena, J. A.; De Alfonso, C. J.

*Chem. Eng. Data* 1986, 31, 339.

**ORTJ0872** – Ortega, J.; Pena, J. A.; De Alfonso, C. *Rev. Latinoam. Ing. Quim. Quim. Apl.* 1987, 17, 195.

**ORTJ0905** – Ortega, J.; Susial, P.; De Alfonso, C. *J. Chem. Eng. Data* 1990, 35, 216.

**ORTJ0906** – Ortega, J.; Susial, P.; De Alfonso, C. *J. Chem. Eng. Data* 1990, 35, 350.

**ORTJ0907** – Ortega, J.; Susial, P. *J. Chem. Eng. Jpn.* 1990, 23, 349.

**ORTJ0934** – Ortega, J.; Susial, P. *J. Chem. Eng. Jpn.* 1993, 26, 259.

**PATV1771** – Patlasov, V. P.; Naumova, T. B.; Sergacheva, N. I.; Balashov, M. I. *Temat. Sb. - Mosk. Inst. Tonk. Khim. Tekhnol. im. D. I. Mendeleeva* 1977, 72), 179.

**PERV0800** – Pereygin, V. M.; Suntsov, Yu. K. *Izv.*

*Vyssh. Uchebn. Zaved., Pishch. Tekhnol.* 1980, (6), 144.

**SUSP0890** – Susial, P.; Ortega, J.; De Alfonso, C.; Alonso, C. *J. Chem. Eng. Data* 1989, 34, 247.

**SUSP0930** – Susial, P.; Ortega, J. *J. Chem. Eng. Data* 1993, 38, 434.

**SUSP0931** – Susial, P.; Ortega, J. *J. Chem. Eng. Data* 1993, 38, 647.

**WICI0930** – Wichterle, I.; Linek, J.; Wagner, Z.; Kehiaian, H. V. *Vapor-Liquid Equilibrium Bibliographic Database, ELDATA, Montreuil, France, 1993.*

.....  
**Ortega, Juan\*** (ORTJ0)

**Susial, Pedro** (SUSP0)

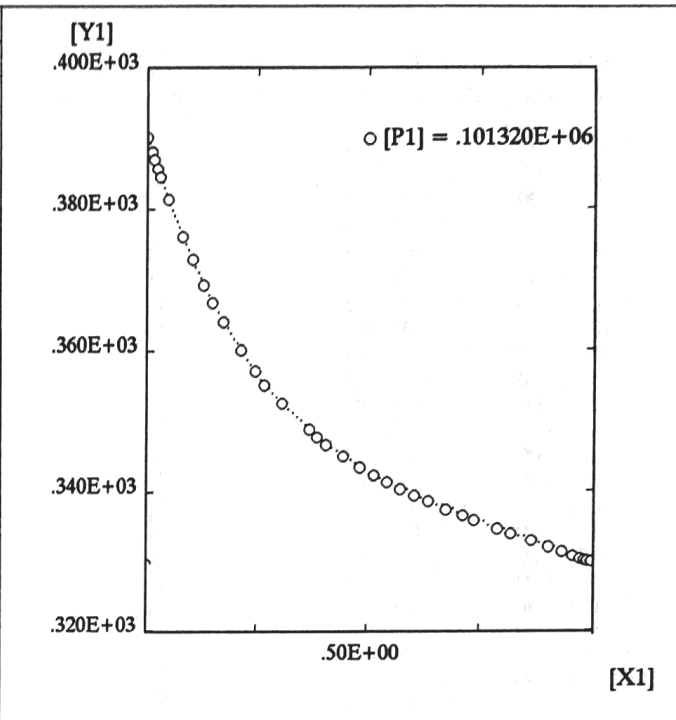
\*Author to whom correspondence should be addressed:

FAX +34-28-451022

**Property Code:** [EVL0001] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS **ORTJ0950.001**  
**State:** Two-component system, single-phase liquid  
 Pure component 1, liquid  
 Pure component 2, liquid  
**Parameters:** [P1] P/Pa, Pressure  
**Variables:** [X1]  $x_1$ /-, Mole fraction of component 1 in liquid phase  
 [Y1] T/K, Temperature  
**Method:** Direct measurement of T at variable  $x_1$  and constant P

**Components:** 1. C<sub>3</sub>H<sub>6</sub>O<sub>2</sub>, Methyl ethanoate  
 2. C<sub>4</sub>H<sub>10</sub>O, 1-Butanol

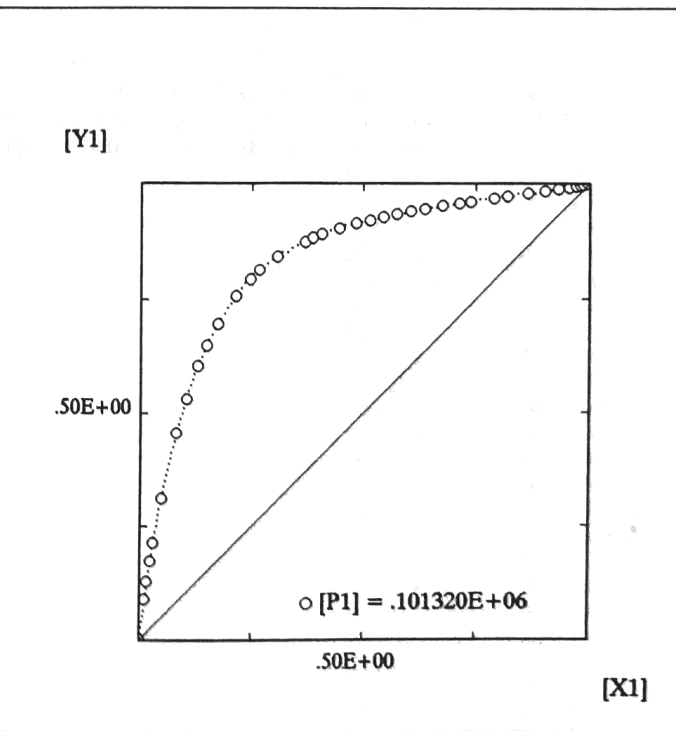
[P1] = .101320E+06			
[X1]	[Y1]	[X1]	[Y1]
.000000E+00	.390230E+03	.678700E+00	.337320E+03
.113000E-01	.388160E+03	.717000E+00	.336540E+03
.169000E-01	.387030E+03	.742000E+00	.335900E+03
.236000E-01	.385790E+03	.792800E+00	.334660E+03
.304000E-01	.384570E+03	.822500E+00	.333930E+03
.491000E-01	.381370E+03	.867200E+00	.332940E+03
.828000E-01	.376010E+03	.905500E+00	.332030E+03
.104800E+00	.372820E+03	.934400E+00	.331320E+03
.131100E+00	.369280E+03	.958100E+00	.330750E+03
.151000E+00	.366790E+03	.974500E+00	.330360E+03
.176400E+00	.364020E+03	.983500E+00	.330150E+03
.215800E+00	.360010E+03	.989400E+00	.330010E+03
.249000E+00	.357080E+03	.100000E+01	.329910E+03
.270300E+00	.355270E+03		
.309400E+00	.352430E+03		
.372700E+00	.348710E+03		
.390100E+00	.347620E+03		
.410500E+00	.346560E+03		
.448800E+00	.344930E+03		
.487000E+00	.343360E+03		
.517500E+00	.342240E+03		
.547400E+00	.341210E+03		
.577200E+00	.340290E+03		
.607900E+00	.339360E+03		
.640300E+00	.338520E+03		



**Property Code:** [EVL0002] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS **ORTJ0950.002**  
**State:** Two-component system, single-phase liquid  
 Pure component 1, liquid  
 Pure component 2, liquid  
**Parameters:** [P1] P/Pa, Pressure  
**Variables:** [X1]  $x_1$ /-, Mole fraction of component 1 in liquid phase  
 [Y1]  $y_1$ /-, Mole fraction of component 1 in vapor phase  
**Method:** Direct measurement of  $y_1$  at variable  $x_1$  and constant P

**Components:** 1. C<sub>3</sub>H<sub>6</sub>O<sub>2</sub>, Methyl ethanoate  
 2. C<sub>4</sub>H<sub>10</sub>O, 1-Butanol

[P1] = .101320E+06			
[X1]	[Y1]	[X1]	[Y1]
.000000E+00	.000000E+00	.678700E+00	.951700E+00
.113000E-01	.883000E-01	.717000E+00	.957000E+00
.169000E-01	.127500E+00	.742000E+00	.960600E+00
.236000E-01	.172500E+00	.792800E+00	.968300E+00
.304000E-01	.212600E+00	.822500E+00	.972400E+00
.491000E-01	.311500E+00	.867200E+00	.978800E+00
.828000E-01	.456700E+00	.905500E+00	.984100E+00
.104800E+00	.531500E+00	.934400E+00	.988500E+00
.131100E+00	.605100E+00	.958100E+00	.991900E+00
.151000E+00	.650900E+00	.974500E+00	.994500E+00
.176400E+00	.697900E+00	.983500E+00	.996100E+00
.215800E+00	.757200E+00	.989400E+00	.997000E+00
.249000E+00	.793100E+00	.100000E+01	.100000E+01
.270300E+00	.813000E+00		
.309400E+00	.840800E+00		
.372700E+00	.873200E+00		
.390100E+00	.881700E+00		
.410500E+00	.889900E+00		
.448800E+00	.902600E+00		
.487000E+00	.913200E+00		
.517500E+00	.920100E+00		
.547400E+00	.927100E+00		
.577200E+00	.933500E+00		
.607900E+00	.939200E+00		
.640300E+00	.944500E+00		





<b>Property Code:</b> [EVL0001] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS				<b>ORTJ0950.003</b>
<b>State:</b> Two-component system, single-phase liquid Pure component 1, liquid Pure component 2, liquid				
<b>Parameters:</b> [P1] P/Pa, Pressure				
<b>Variables:</b> [X1] $x_1$ /-, Mole fraction of component 1 in liquid phase [Y1] T/K, Temperature				
<b>Method:</b> Direct measurement of T at variable $x_1$ and constant P				
<b>Components:</b> 1. C <sub>3</sub> H <sub>6</sub> O <sub>2</sub> , Methyl ethanoate 2. C <sub>4</sub> H <sub>10</sub> O, 2-Butanol				
[P1] = .101320E+06				
[X1]	[Y1]	[X1]	[Y1]	
.000000E+00	.372360E+03	.942300E+00	.330920E+03	
.213000E-01	.370280E+03	.969900E+00	.330340E+03	
.270000E-01	.369800E+03	.100000E+01	.329910E+03	
.571000E-01	.367090E+03			
.136300E+00	.360400E+03			
.157400E+00	.358300E+03			
.180500E+00	.356580E+03			
.202900E+00	.355070E+03			
.225400E+00	.353690E+03			
.244400E+00	.352400E+03			
.267900E+00	.351110E+03			
.296800E+00	.349480E+03			
.320500E+00	.348330E+03			
.343000E+00	.347280E+03			
.366900E+00	.346300E+03			
.402300E+00	.345030E+03			
.423200E+00	.344530E+03			
.462300E+00	.343040E+03			
.513900E+00	.342050E+03			
.610900E+00	.338700E+03			
.673800E+00	.337370E+03			
.760100E+00	.335180E+03			
.806600E+00	.333890E+03			
.857300E+00	.332810E+03			
.898800E+00	.331900E+03			

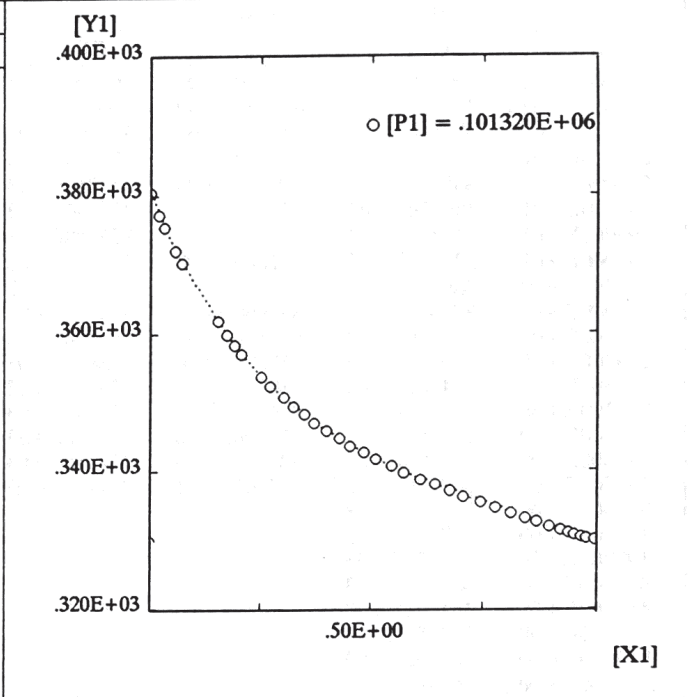
<b>Property Code:</b> [EVL0002] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS				<b>ORTJ0950.004</b>
<b>State:</b> Two-component system, single-phase liquid Pure component 1, liquid Pure component 2, liquid				
<b>Parameters:</b> [P1] P/Pa, Pressure				
<b>Variables:</b> [X1] $x_1$ /-, Mole fraction of component 1 in liquid phase [Y1] $y_1$ /-, Mole fraction of component 1 in vapor phase				
<b>Method:</b> Direct measurement of $y_1$ at variable $x_1$ and constant P				
<b>Components:</b> 1. C <sub>3</sub> H <sub>6</sub> O <sub>2</sub> , Methyl ethanoate 2. C <sub>4</sub> H <sub>10</sub> O, 2-Butanol				
[P1] = .101320E+06				
[X1]	[Y1]	[X1]	[Y1]	
.000000E+00	.000000E+00	.942300E+00	.982900E+00	
.213000E-01	.987000E-01	.969900E+00	.990800E+00	
.270000E-01	.120000E+00	.100000E+01	.100000E+01	
.571000E-01	.232700E+00			
.136300E+00	.445300E+00			
.157400E+00	.496000E+00			
.180500E+00	.540700E+00			
.202900E+00	.578300E+00			
.225400E+00	.609000E+00			
.244400E+00	.637500E+00			
.267900E+00	.665300E+00			
.296800E+00	.696100E+00			
.320500E+00	.718000E+00			
.343000E+00	.737100E+00			
.366900E+00	.754700E+00			
.402300E+00	.777700E+00			
.423200E+00	.786800E+00			
.462300E+00	.811900E+00			
.513900E+00	.831500E+00			
.610900E+00	.879100E+00			
.673800E+00	.894700E+00			
.760100E+00	.923700E+00			
.806600E+00	.941600E+00			
.857300E+00	.956300E+00			
.898800E+00	.969500E+00			

**Property Code:** [EVLMO001] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS ORTJ0950.005  
**State:** Two-component system, single-phase liquid  
 Pure component 1, liquid  
 Pure component 2, liquid  
**Parameters:** [P1] P/Pa, Pressure  
**Variables:** [X1]  $x_1$ /-, Mole fraction of component 1 in liquid phase  
 [Y1] T/K, Temperature  
**Method:** Direct measurement of  $T$  at variable  $x_1$  and constant  $P$

**Components:** 1. C<sub>3</sub>H<sub>6</sub>O<sub>2</sub>, Methyl ethanoate  
 2. C<sub>4</sub>H<sub>10</sub>O, 2-Methyl-1-propanol

[P1] = .101320E+06

[X1]	[Y1]	[X1]	[Y1]
.000000E+00	.380390E+03	.708500E+00	.336190E+03
.191000E-01	.377180E+03	.747300E+00	.335380E+03
.316000E-01	.375400E+03	.780500E+00	.334560E+03
.572000E-01	.371960E+03	.815300E+00	.333760E+03
.723000E-01	.370120E+03	.845500E+00	.333090E+03
.155000E+00	.361810E+03	.871200E+00	.332520E+03
.176500E+00	.359830E+03	.898800E+00	.331880E+03
.193600E+00	.358340E+03	.923700E+00	.331370E+03
.209100E+00	.357050E+03	.941500E+00	.330970E+03
.253700E+00	.353790E+03	.954500E+00	.330690E+03
.274700E+00	.352330E+03	.969600E+00	.330350E+03
.306000E+00	.350700E+03	.980300E+00	.330100E+03
.327100E+00	.349330E+03	.100000E+01	.329910E+03
.351800E+00	.348220E+03		
.374000E+00	.346980E+03		
.402100E+00	.345860E+03		
.432600E+00	.344710E+03		
.455600E+00	.343580E+03		
.486100E+00	.342600E+03		
.513300E+00	.341620E+03		
.549500E+00	.340630E+03		
.575300E+00	.339650E+03		
.612900E+00	.338680E+03		
.645900E+00	.337930E+03		
.678500E+00	.337000E+03		

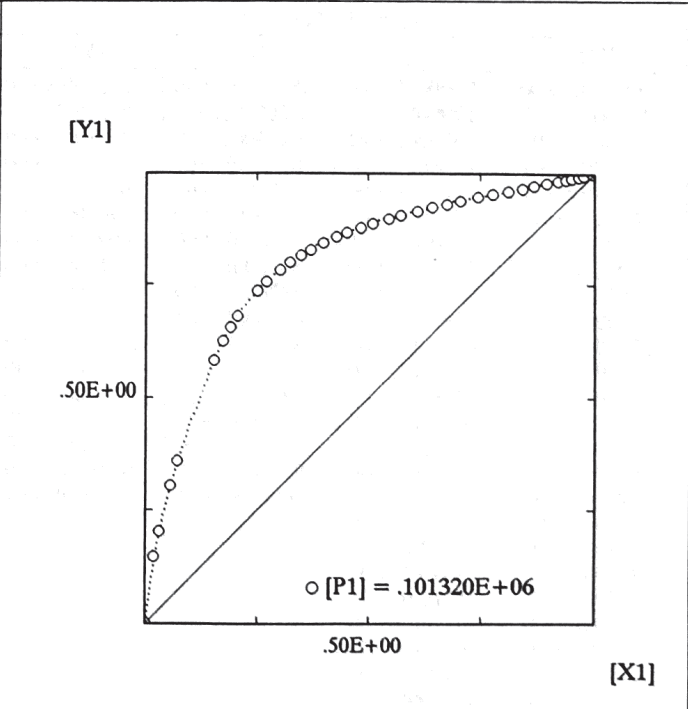


**Property Code:** [EVLMO002] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS ORTJ0950.006  
**State:** Two-component system, single-phase liquid  
 Pure component 1, liquid  
 Pure component 2, liquid  
**Parameters:** [P1] P/Pa, Pressure  
**Variables:** [X1]  $x_1$ /-, Mole fraction of component 1 in liquid phase  
 [Y1]  $y_1$ /-, Mole fraction of component 1 in vapor phase  
**Method:** Direct measurement of  $y_1$  at variable  $x_1$  and constant  $P$

**Components:** 1. C<sub>3</sub>H<sub>6</sub>O<sub>2</sub>, Methyl ethanoate  
 2. C<sub>4</sub>H<sub>10</sub>O, 2-Methyl-1-propanol

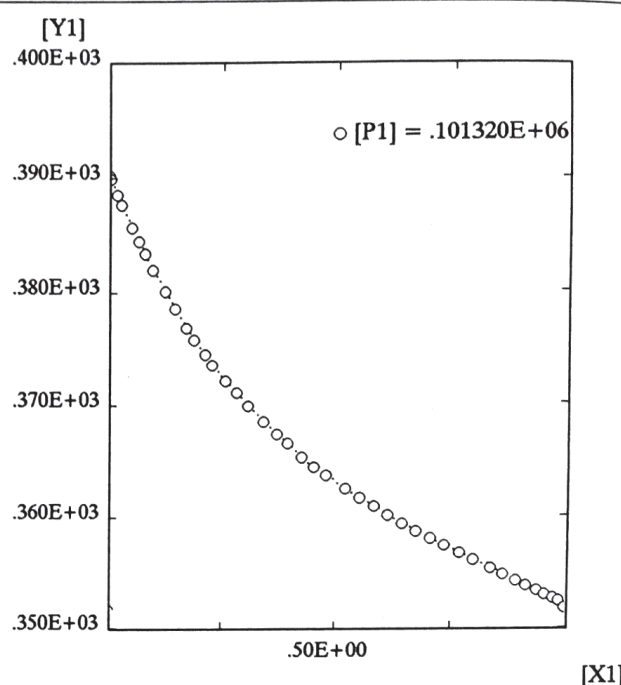
[P1] = .101320E+06

[X1]	[Y1]	[X1]	[Y1]
.000000E+00	.000000E+00	.708500E+00	.938200E+00
.191000E-01	.146300E+00	.747300E+00	.946900E+00
.316000E-01	.201700E+00	.780500E+00	.953500E+00
.572000E-01	.303600E+00	.815300E+00	.960800E+00
.723000E-01	.358600E+00	.845500E+00	.966900E+00
.155000E+00	.581700E+00	.871200E+00	.972100E+00
.176500E+00	.624700E+00	.898800E+00	.978300E+00
.193600E+00	.653800E+00	.923700E+00	.983300E+00
.209100E+00	.678600E+00	.941500E+00	.987000E+00
.253700E+00	.734600E+00	.954500E+00	.989900E+00
.274700E+00	.755000E+00	.969600E+00	.993400E+00
.306000E+00	.781800E+00	.980300E+00	.995700E+00
.327100E+00	.799500E+00	.100000E+01	.100000E+01
.351800E+00	.814400E+00		
.374000E+00	.827800E+00		
.402100E+00	.843000E+00		
.432600E+00	.856700E+00		
.455600E+00	.866300E+00		
.486100E+00	.877500E+00		
.513300E+00	.887000E+00		
.549500E+00	.897200E+00		
.575300E+00	.904900E+00		
.612900E+00	.914400E+00		
.645900E+00	.923600E+00		
.678500E+00	.930700E+00		

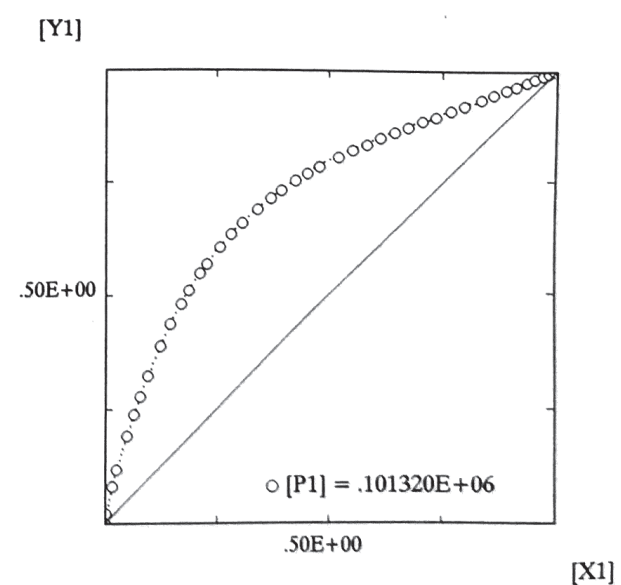




<b>Property Code:</b> [EVL0001] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS		<b>ORTJ0950.007</b>	
<b>State:</b> Two-component system, single-phase liquid Pure component 1, liquid Pure component 2, liquid			
<b>Parameters:</b> [P1] P/Pa, Pressure			
<b>Variables:</b> [X1] $x_1$ /-, Mole fraction of component 1 in liquid phase [Y1] T/K, Temperature			
<b>Method:</b> Direct measurement of T at variable $x_1$ and constant P			
<b>Components:</b> 1. C <sub>4</sub> H <sub>8</sub> O <sub>2</sub> , Methyl propanoate 2. C <sub>4</sub> H <sub>10</sub> O, 1-Butanol			
[P1] = .101320E+06			
[X1]	[Y1]	[X1]	[Y1]
.000000E+00	.390230E+03	.586100E+00	.360900E+03
.440000E-02	.389920E+03	.616500E+00	.360070E+03
.182000E-01	.388560E+03	.648300E+00	.359350E+03
.280000E-01	.387650E+03	.677900E+00	.358680E+03
.513000E-01	.385670E+03	.709200E+00	.358010E+03
.662000E-01	.384450E+03	.737900E+00	.357380E+03
.799000E-01	.383380E+03	.771600E+00	.356690E+03
.969000E-01	.381960E+03	.801200E+00	.356100E+03
.126300E+00	.380120E+03	.837600E+00	.355320E+03
.148000E+00	.378620E+03	.865600E+00	.354770E+03
.172500E+00	.376880E+03	.893100E+00	.354230E+03
.189800E+00	.375860E+03	.915000E+00	.353800E+03
.214600E+00	.374560E+03	.938500E+00	.353320E+03
.229700E+00	.373620E+03	.956000E+00	.352990E+03
.259800E+00	.372180E+03	.973800E+00	.352660E+03
.284600E+00	.371090E+03	.987200E+00	.352410E+03
.309700E+00	.369900E+03	.100000E+01	.351770E+03
.344000E+00	.368450E+03		
.373700E+00	.367350E+03		
.397200E+00	.366530E+03		
.428300E+00	.365300E+03		
.454800E+00	.364430E+03		
.481600E+00	.363650E+03		
.523500E+00	.362460E+03		
.554800E+00	.361660E+03		



<b>Property Code:</b> [EVL0002] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS		<b>ORTJ0950.008</b>	
<b>State:</b> Two-component system, single-phase liquid Pure component 1, liquid Pure component 2, liquid			
<b>Parameters:</b> [P1] P/Pa, Pressure			
<b>Variables:</b> [X1] $x_1$ /-, Mole fraction of component 1 in liquid phase [Y1] $y_1$ /-, Mole fraction of component 1 in vapor phase			
<b>Method:</b> Direct measurement of $y_1$ at variable $x_1$ and constant P			
<b>Components:</b> 1. C <sub>4</sub> H <sub>8</sub> O <sub>2</sub> , Methyl propanoate 2. C <sub>4</sub> H <sub>10</sub> O, 1-Butanol			
[P1] = .101320E+06			
[X1]	[Y1]	[X1]	[Y1]
.000000E+00	.000000E+00	.586100E+00	.833800E+00
.440000E-02	.199000E-01	.616500E+00	.848600E+00
.182000E-01	.787000E-01	.648300E+00	.861100E+00
.280000E-01	.116800E+00	.677900E+00	.873200E+00
.513000E-01	.192000E+00	.709200E+00	.886200E+00
.662000E-01	.238200E+00	.737900E+00	.896800E+00
.799000E-01	.278000E+00	.771600E+00	.909800E+00
.969000E-01	.323100E+00	.801200E+00	.921400E+00
.126300E+00	.388800E+00	.837600E+00	.934900E+00
.148000E+00	.436700E+00	.865600E+00	.945700E+00
.172500E+00	.480200E+00	.893100E+00	.956300E+00
.189800E+00	.510400E+00	.915000E+00	.965500E+00
.214600E+00	.547800E+00	.938500E+00	.974600E+00
.229700E+00	.567800E+00	.956000E+00	.981800E+00
.259800E+00	.604800E+00	.973800E+00	.989300E+00
.284600E+00	.633700E+00	.987200E+00	.994800E+00
.309700E+00	.658900E+00	.100000E+01	.100000E+01
.344000E+00	.688200E+00		
.373700E+00	.712820E+00		
.397200E+00	.730400E+00		
.428300E+00	.753100E+00		
.454800E+00	.768000E+00		
.481600E+00	.783800E+00		
.523500E+00	.804900E+00		
.554800E+00	.820500E+00		

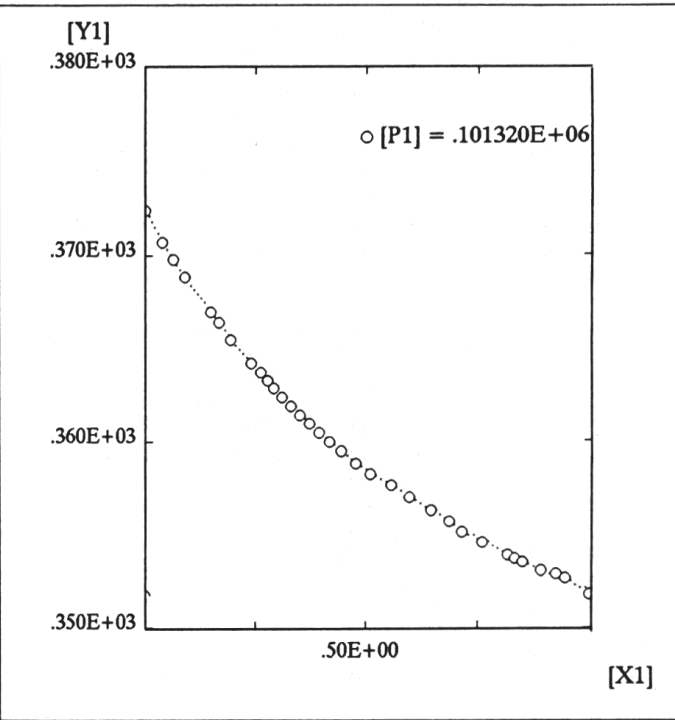


**Property Code:** [EVL0001] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS ORTJ0950.009  
**State:** Two-component system, single-phase liquid  
 Pure component 1, liquid  
 Pure component 2, liquid  
**Parameters:** [P1] P/Pa, Pressure  
**Variables:** [X1]  $x_1$ /-, Mole fraction of component 1 in liquid phase  
 [Y1] T/K, Temperature  
**Method:** Direct measurement of T at variable  $x_1$  and constant P

**Components:** 1. C<sub>4</sub>H<sub>8</sub>O<sub>2</sub>, Methyl propanoate  
 2. C<sub>4</sub>H<sub>10</sub>O, 2-Butanol

[P1] = .101320E+06

[X1]	[Y1]	[X1]	[Y1]
.000000E+00	.372360E+03	.762900E+00	.354570E+03
.397000E-01	.370680E+03	.819900E+00	.353870E+03
.644000E-01	.369800E+03	.835500E+00	.353680E+03
.910000E-01	.368840E+03	.853000E+00	.353520E+03
.151100E+00	.366980E+03	.893000E+00	.353060E+03
.169300E+00	.366390E+03	.925200E+00	.352860E+03
.196000E+00	.365450E+03	.946400E+00	.352660E+03
.243500E+00	.364190E+03	.100000E+01	.351770E+03
.265600E+00	.363700E+03		
.279900E+00	.363270E+03		
.294000E+00	.362860E+03		
.313800E+00	.362340E+03		
.333400E+00	.361850E+03		
.353800E+00	.361380E+03		
.375400E+00	.360920E+03		
.397800E+00	.360430E+03		
.420900E+00	.359970E+03		
.446900E+00	.359450E+03		
.479800E+00	.358810E+03		
.513700E+00	.358230E+03		
.560500E+00	.357630E+03		
.600300E+00	.356990E+03		
.648900E+00	.356270E+03		
.690100E+00	.355690E+03		
.717200E+00	.355120E+03		

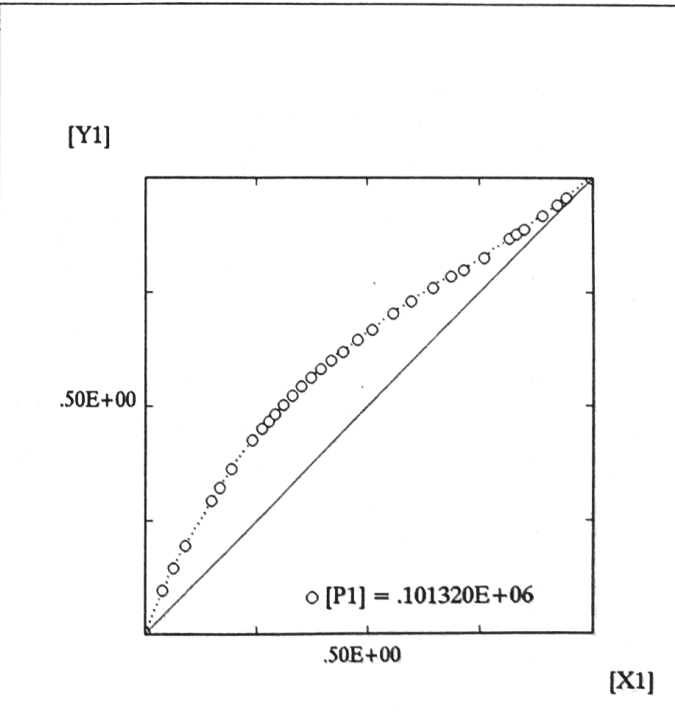


**Property Code:** [EVL0002] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS ORTJ0950.010  
**State:** Two-component system, single-phase liquid  
 Pure component 1, liquid  
 Pure component 2, liquid  
**Parameters:** [P1] P/Pa, Pressure  
**Variables:** [X1]  $x_1$ /-, Mole fraction of component 1 in liquid phase  
 [Y1]  $y_1$ /-, Mole fraction of component 1 in vapor phase  
**Method:** Direct measurement of  $y_1$  at variable  $x_1$  and constant P

**Components:** 1. C<sub>4</sub>H<sub>8</sub>O<sub>2</sub>, Methyl propanoate  
 2. C<sub>4</sub>H<sub>10</sub>O, 2-Butanol

[P1] = .101320E+06

[X1]	[Y1]	[X1]	[Y1]
.000000E+00	.000000E+00	.762900E+00	.824200E+00
.397000E-01	.941000E-01	.819900E+00	.866000E+00
.644000E-01	.144300E+00	.835500E+00	.876300E+00
.910000E-01	.195300E+00	.853000E+00	.887300E+00
.151100E+00	.295100E+00	.893000E+00	.915800E+00
.169300E+00	.323300E+00	.925200E+00	.939700E+00
.196000E+00	.363800E+00	.946400E+00	.956100E+00
.243500E+00	.425900E+00	.100000E+01	.100000E+01
.265600E+00	.450400E+00		
.279900E+00	.465900E+00		
.294000E+00	.481300E+00		
.313800E+00	.502400E+00		
.333400E+00	.523100E+00		
.353800E+00	.542400E+00		
.375400E+00	.561500E+00		
.397800E+00	.580300E+00		
.420900E+00	.599200E+00		
.446900E+00	.618700E+00		
.479800E+00	.644600E+00		
.513700E+00	.666600E+00		
.560500E+00	.702500E+00		
.600300E+00	.728300E+00		
.648900E+00	.757900E+00		
.690100E+00	.784100E+00		
.717200E+00	.798200E+00		





<b>Property Code:</b> [EVL0001] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS				<b>ORTJ0950.011</b>
<b>State:</b> Two-component system, single-phase liquid Pure component 1, liquid Pure component 2, liquid				
<b>Parameters:</b> [P1] P/Pa, Pressure				
<b>Variables:</b> [X1] x <sub>1</sub> /-, Mole fraction of component 1 in liquid phase [Y1] T/K, Temperature				
<b>Method:</b> Direct measurement of T at variable x <sub>1</sub> and constant P				
<b>Components:</b> 1. C <sub>4</sub> H <sub>8</sub> O <sub>2</sub> , Methyl propanoate 2. C <sub>4</sub> H <sub>10</sub> O, 2-Methyl-1-propanol				
[P1] = .101320E+06				
[X1]	[Y1]	[X1]	[Y1]	
.000000E+00	.380390E+03	.607300E+00	.358410E+03	
.214000E-01	.378710E+03	.635700E+00	.357910E+03	
.362000E-01	.377830E+03	.664200E+00	.357420E+03	
.534000E-01	.376870E+03	.694700E+00	.356830E+03	
.690000E-01	.376080E+03	.719100E+00	.356410E+03	
.840000E-01	.375270E+03	.756000E+00	.355760E+03	
.997000E-01	.374500E+03	.787700E+00	.355230E+03	
.115500E+00	.373700E+03	.821000E+00	.354770E+03	
.134700E+00	.372850E+03	.854300E+00	.354170E+03	
.153500E+00	.371970E+03	.886000E+00	.353760E+03	
.173600E+00	.371080E+03	.912500E+00	.353380E+03	
.193500E+00	.370220E+03	.940900E+00	.352920E+03	
.217400E+00	.369300E+03	.968900E+00	.352460E+03	
.236200E+00	.368560E+03	.984400E+00	.352200E+03	
.279400E+00	.366780E+03	.100000E+01	.351770E+03	
.308900E+00	.365720E+03			
.338600E+00	.364820E+03			
.367300E+00	.363910E+03			
.395500E+00	.363140E+03			
.423900E+00	.362410E+03			
.453200E+00	.361740E+03			
.482700E+00	.360980E+03			
.507500E+00	.360510E+03			
.542700E+00	.359720E+03			
.575700E+00	.359030E+03			

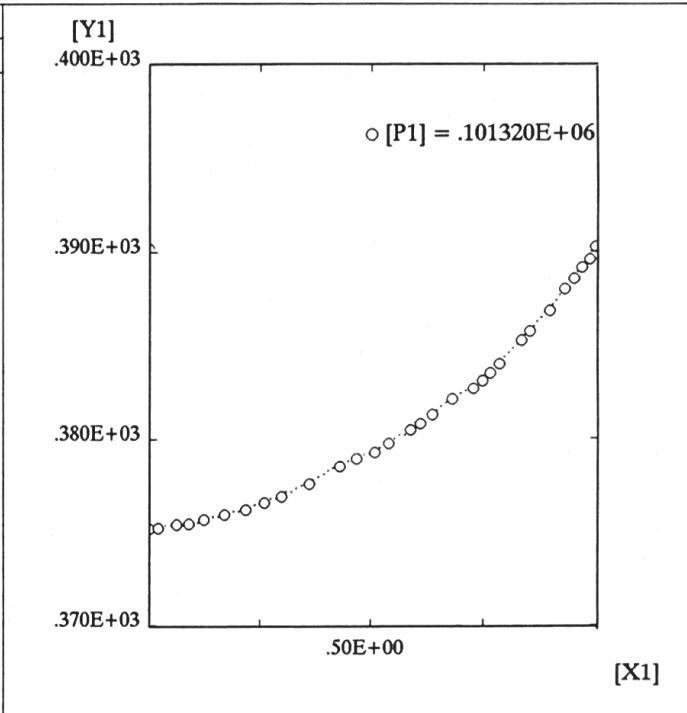
<b>Property Code:</b> [EVL0002] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS				<b>ORTJ0950.012</b>
<b>State:</b> Two-component system, single-phase liquid Pure component 1, liquid Pure component 2, liquid				
<b>Parameters:</b> [P1] P/Pa, Pressure				
<b>Variables:</b> [X1] x <sub>1</sub> /-, Mole fraction of component 1 in liquid phase [Y1] y <sub>1</sub> /-, Mole fraction of component 1 in vapor phase				
<b>Method:</b> Direct measurement of y <sub>1</sub> at variable x <sub>1</sub> and constant P				
<b>Components:</b> 1. C <sub>4</sub> H <sub>8</sub> O <sub>2</sub> , Methyl propanoate 2. C <sub>4</sub> H <sub>10</sub> O, 2-Methyl-1-propanol				
[P1] = .101320E+06				
[X1]	[Y1]	[X1]	[Y1]	
.000000E+00	.000000E+00	.607300E+00	.804300E+00	
.214000E-01	.884000E-01	.635700E+00	.819200E+00	
.362000E-01	.127900E+00	.664200E+00	.835000E+00	
.534000E-01	.172000E+00	.694700E+00	.850100E+00	
.690000E-01	.208100E+00	.719100E+00	.860700E+00	
.840000E-01	.243400E+00	.756000E+00	.890400E+00	
.997000E-01	.275500E+00	.787700E+00	.896000E+00	
.115500E+00	.309000E+00	.821000E+00	.912500E+00	
.134700E+00	.344900E+00	.854300E+00	.929100E+00	
.153500E+00	.377000E+00	.886000E+00	.944600E+00	
.173600E+00	.410000E+00	.912500E+00	.958100E+00	
.193500E+00	.441200E+00	.940900E+00	.972000E+00	
.217400E+00	.475000E+00	.968900E+00	.985300E+00	
.236200E+00	.499900E+00	.984400E+00	.992600E+00	
.279400E+00	.557000E+00	.100000E+01	.100000E+01	
.308900E+00	.589000E+00			
.338600E+00	.617700E+00			
.367300E+00	.645300E+00			
.395500E+00	.666400E+00			
.423900E+00	.689000E+00			
.453200E+00	.710300E+00			
.482700E+00	.730000E+00			
.507500E+00	.746800E+00			
.542700E+00	.767900E+00			
.575700E+00	.787900E+00			



**Property Code:** [EVL0001] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS ORTJ0950.013  
**State:** Two-component system, single-phase liquid  
 Pure component 1, liquid  
 Pure component 2, liquid  
**Parameters:** [P1] P/Pa, Pressure  
**Variables:** [X1]  $x_1$ /-, Mole fraction of component 1 in liquid phase  
 [Y1] T/K, Temperature  
**Method:** Direct measurement of T at variable  $x_1$  and constant P

**Components:** 1. C<sub>4</sub>H<sub>10</sub>O, 1-Butanol  
 2. C<sub>5</sub>H<sub>10</sub>O<sub>2</sub>, Methyl butanoate

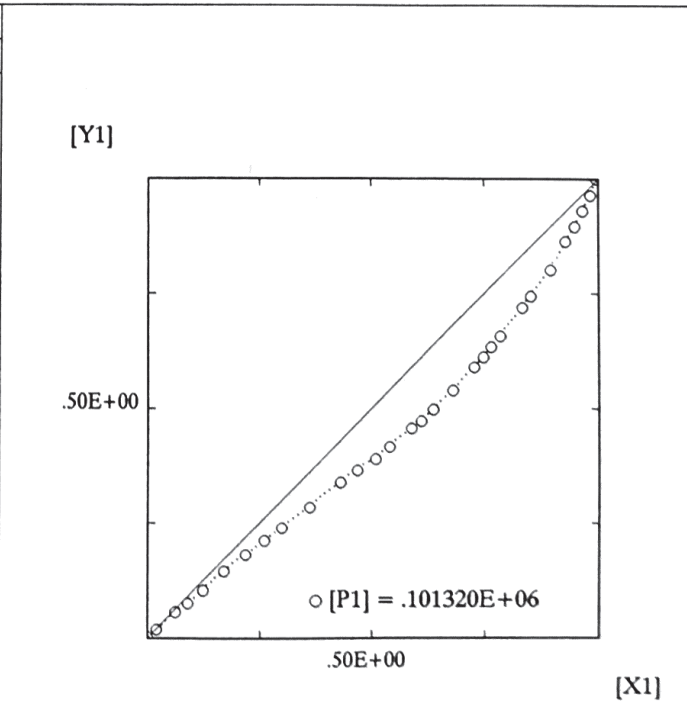
[P1] = .101320E+06			
[X1]	[Y1]	[X1]	[Y1]
.000000E+00	.375150E+03	.931600E+00	.387930E+03
.206000E-01	.375180E+03	.952100E+00	.388500E+03
.635000E-01	.375390E+03	.968900E+00	.389120E+03
.913000E-01	.375420E+03	.987400E+00	.389550E+03
.125800E+00	.375640E+03	.100000E+01	.390230E+03
.173200E+00	.375890E+03		
.221000E+00	.376160E+03		
.263500E+00	.376540E+03		
.302800E+00	.376870E+03		
.364600E+00	.377560E+03		
.434300E+00	.378490E+03		
.471500E+00	.378920E+03		
.511800E+00	.379240E+03		
.542200E+00	.379740E+03		
.591400E+00	.380450E+03		
.613000E+00	.380800E+03		
.639800E+00	.381280E+03		
.683100E+00	.382110E+03		
.729800E+00	.382660E+03		
.751300E+00	.383080E+03		
.768700E+00	.383480E+03		
.788800E+00	.383980E+03		
.836700E+00	.385190E+03		
.855600E+00	.385710E+03		
.898500E+00	.386770E+03		



**Property Code:** [EVL0002] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS ORTJ0950.014  
**State:** Two-component system, single-phase liquid  
 Pure component 1, liquid  
 Pure component 2, liquid  
**Parameters:** [P1] P/Pa, Pressure  
**Variables:** [X1]  $x_1$ /-, Mole fraction of component 1 in liquid phase  
 [Y1]  $y_1$ /-, Mole fraction of component 1 in vapor phase  
**Method:** Direct measurement of  $y_1$  at variable  $x_1$  and constant P

**Components:** 1. C<sub>4</sub>H<sub>10</sub>O, 1-Butanol  
 2. C<sub>5</sub>H<sub>10</sub>O<sub>2</sub>, Methyl butanoate

[P1] = .101320E+06			
[X1]	[Y1]	[X1]	[Y1]
.000000E+00	.000000E+00	.931600E+00	.863500E+00
.206000E-01	.166000E-01	.952100E+00	.895700E+00
.635000E-01	.537000E-01	.968900E+00	.930300E+00
.913000E-01	.725000E-01	.987400E+00	.965100E+00
.125800E+00	.100600E+00	.100000E+01	.100000E+01
.173200E+00	.143600E+00		
.221000E+00	.178400E+00		
.263500E+00	.209700E+00		
.302800E+00	.238500E+00		
.364600E+00	.283300E+00		
.434300E+00	.337900E+00		
.471500E+00	.364800E+00		
.511800E+00	.390100E+00		
.542200E+00	.416000E+00		
.591400E+00	.456900E+00		
.613000E+00	.473100E+00		
.639800E+00	.498100E+00		
.683100E+00	.537800E+00		
.729800E+00	.587600E+00		
.751300E+00	.610700E+00		
.768700E+00	.632300E+00		
.788800E+00	.656100E+00		
.836700E+00	.717200E+00		
.855600E+00	.743700E+00		
.898500E+00	.800300E+00		



Property Code: [EVL0001] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS				ORTJ0950.015
<b>State:</b> Two-component system, single-phase liquid				
Pure component 1, liquid				
Pure component 2, liquid				
<b>Parameters:</b> [P1] P/Pa, Pressure				
<b>Variables:</b> [X1] $x_1$ /-, Mole fraction of component 1 in liquid phase				
[Y1] T/K, Temperature				
<b>Method:</b> Direct measurement of T at variable $x_1$ and constant P				
<b>Components:</b> 1. C <sub>4</sub> H <sub>10</sub> O, 2-Butanol 2. C <sub>5</sub> H <sub>10</sub> O <sub>2</sub> , Methyl butanoate				
[P1] = .101320E+06				
[X1]	[Y1]	[X1]	[Y1]	
.000000E+00	.375150E+03	.820100E+00	.370980E+03	
.830001E-02	.375110E+03	.864200E+00	.371220E+03	
.194000E-01	.375070E+03	.890300E+00	.371390E+03	
.382000E-01	.374770E+03	.914500E+00	.371540E+03	
.701000E-01	.374230E+03	.935500E+00	.371660E+03	
.100800E+00	.373730E+03	.966700E+00	.371950E+03	
.133400E+00	.373350E+03	.979900E+00	.372130E+03	
.165100E+00	.372970E+03	.992900E+00	.372360E+03	
.196900E+00	.372640E+03	.100000E+01	.372360E+03	
.225400E+00	.372390E+03			
.297500E+00	.371810E+03			
.339500E+00	.371530E+03			
.368400E+00	.371390E+03			
.457200E+00	.370800E+03			
.506200E+00	.370770E+03			
.538000E+00	.370700E+03			
.569300E+00	.370660E+03			
.599500E+00	.370630E+03			
.627200E+00	.370630E+03			
.655100E+00	.370640E+03			
.686500E+00	.370660E+03			
.712000E+00	.370690E+03			
.738500E+00	.370730E+03			
.766900E+00	.370810E+03			
.796600E+00	.370900E+03			

Property Code: [EVL0002] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS				ORTJ0950.016
<b>State:</b> Two-component system, single-phase liquid				
Pure component 1, liquid				
Pure component 2, liquid				
<b>Parameters:</b> [P1] P/Pa, Pressure				
<b>Variables:</b> [X1] $x_1$ /-, Mole fraction of component 1 in liquid phase				
[Y1] $y_1$ /-, Mole fraction of component 1 in vapor phase				
<b>Method:</b> Direct measurement of $y_1$ at variable $x_1$ and constant P				
<b>Components:</b> 1. C <sub>4</sub> H <sub>10</sub> O, 2-Butanol 2. C <sub>5</sub> H <sub>10</sub> O <sub>2</sub> , Methyl butanoate				
[P1] = .101320E+06				
[X1]	[Y1]	[X1]	[Y1]	
.000000E+00	.000000E+00	.820100E+00	.792100E+00	
.830001E-02	.136000E-01	.864200E+00	.838500E+00	
.194000E-01	.304000E-01	.890300E+00	.865800E+00	
.382000E-01	.594000E-01	.914500E+00	.893500E+00	
.701000E-01	.109600E+00	.935500E+00	.915800E+00	
.100800E+00	.154200E+00	.966700E+00	.954900E+00	
.133400E+00	.194000E+00	.979900E+00	.972200E+00	
.165100E+00	.231600E+00	.992900E+00	.989300E+00	
.196900E+00	.265100E+00	.100000E+01	.100000E+01	
.225400E+00	.293600E+00			
.297500E+00	.360800E+00			
.339500E+00	.400800E+00			
.368400E+00	.421800E+00			
.457200E+00	.492400E+00			
.506200E+00	.526800E+00			
.538000E+00	.553800E+00			
.569300E+00	.576100E+00			
.599500E+00	.598000E+00			
.627200E+00	.621600E+00			
.655100E+00	.645300E+00			
.686500E+00	.670600E+00			
.712000E+00	.692600E+00			
.738500E+00	.716000E+00			
.766900E+00	.741700E+00			
.796600E+00	.768900E+00			



**Property Code:** [EVL0001] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS ORTJ0950.017  
**State:** Two-component system, single-phase liquid  
 Pure component 1, liquid  
 Pure component 2, liquid  
**Parameters:** [P1] P/Pa, Pressure  
**Variables:** [X1]  $x_1$ /-, Mole fraction of component 1 in liquid phase  
 [Y1] T/K, Temperature  
**Method:** Direct measurement of T at variable  $x_1$  and constant P

**Components:** 1. C<sub>4</sub>H<sub>10</sub>O, 2-Methyl-1-propanol  
 2. C<sub>5</sub>H<sub>10</sub>O<sub>2</sub>, Methyl butanoate

[P1] = .101320E+06			
[X1]	[Y1]	[X1]	[Y1]
.000000E+00	.375150E+03	.802400E+00	.377510E+03
.180000E-01	.374990E+03	.824200E+00	.377820E+03
.403000E-01	.374850E+03	.851000E+00	.378110E+03
.754000E-01	.374700E+03	.879700E+00	.378580E+03
.112200E+00	.374550E+03	.901600E+00	.379010E+03
.147600E+00	.374500E+03	.929600E+00	.379380E+03
.181700E+00	.374400E+03	.952700E+00	.379750E+03
.212000E+00	.374350E+03	.971900E+00	.380240E+03
.246300E+00	.374320E+03	.989300E+00	.380370E+03
.277800E+00	.374330E+03	.100000E+01	.380390E+03
.307900E+00	.374350E+03		
.339900E+00	.374400E+03		
.375900E+00	.374450E+03		
.402300E+00	.374540E+03		
.431300E+00	.374640E+03		
.458100E+00	.374730E+03		
.486600E+00	.374870E+03		
.515400E+00	.375020E+03		
.542400E+00	.375170E+03		
.572400E+00	.375310E+03		
.594600E+00	.375480E+03		
.618000E+00	.375650E+03		
.662500E+00	.376020E+03		
.746400E+00	.376870E+03		
.775700E+00	.377180E+03		

**Property Code:** [EVL0002] VAPOR-LIQUID EQUILIBRIUM IN MIXTURES AND SOLUTIONS ORTJ0950.018  
**State:** Two-component system, single-phase liquid  
 Pure component 1, liquid  
 Pure component 2, liquid  
**Parameters:** [P1] P/Pa, Pressure  
**Variables:** [X1]  $x_1$ /-, Mole fraction of component 1 in liquid phase  
 [Y1]  $y_1$ /-, Mole fraction of component 1 in vapor phase  
**Method:** Direct measurement of  $y_1$  at variable  $x_1$  and constant P

**Components:** 1. C<sub>4</sub>H<sub>10</sub>O, 2-Methyl-1-propanol  
 2. C<sub>5</sub>H<sub>10</sub>O<sub>2</sub>, Methyl butanoate

[P1] = .101320E+06			
[X1]	[Y1]	[X1]	[Y1]
.000000E+00	.000000E+00	.802400E+00	.721400E+00
.180000E-01	.209000E-01	.824200E+00	.748300E+00
.403000E-01	.468000E-01	.851000E+00	.779600E+00
.754000E-01	.863000E-01	.879700E+00	.814500E+00
.112200E+00	.123800E+00	.901600E+00	.849900E+00
.147600E+00	.157800E+00	.929600E+00	.887700E+00
.181700E+00	.191700E+00	.952700E+00	.918800E+00
.212000E+00	.220500E+00	.971900E+00	.954000E+00
.246300E+00	.251500E+00	.989300E+00	.978200E+00
.277800E+00	.278000E+00	.100000E+01	.100000E+01
.307900E+00	.305500E+00		
.339900E+00	.331800E+00		
.375900E+00	.356800E+00		
.402300E+00	.379000E+00		
.431300E+00	.402300E+00		
.458100E+00	.424800E+00		
.486600E+00	.449600E+00		
.515400E+00	.472800E+00		
.542400E+00	.494400E+00		
.572400E+00	.514900E+00		
.594600E+00	.535100E+00		
.618000E+00	.555300E+00		
.662500E+00	.593500E+00		
.746400E+00	.664200E+00		
.775700E+00	.693100E+00		