



**Pedro P. Socorro-Perdomo<sup>1</sup>, Néstor R. Florido-Suárez<sup>1</sup>, Julia C. Mirza-Rosca<sup>1</sup>, Adriana Saceleanu<sup>2</sup>**  
<sup>1</sup> University of Las Palmas de Gran Canaria, Mechanical Engineering Dept.,  
University Campus of Tafira, Engineering Building, 35017 Las Palmas de Gran Canaria, Spain, e-mail: nestor.florido@ulpgc.es  
<sup>2</sup> Lucian Blaga University of Sibiu, Medicine Faculty, 550024 Sibiu, Romania, email: adriana.saceleanu@ulbsibiu.ro

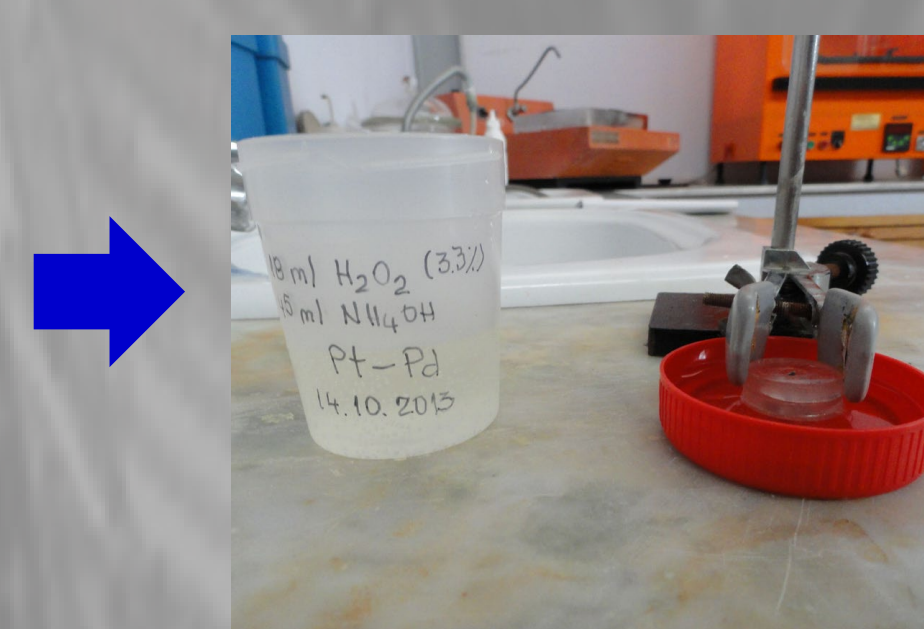


## CORROSION BEHAVIOR OF SOME NI-CR AND CO-CR DENTAL ALLOYS

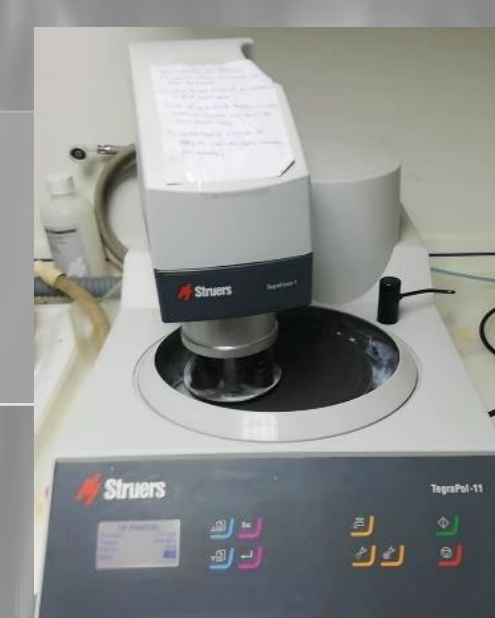
The use of dental alloys as a material for bridges, crowns and prostheses is currently being investigated, taking into account their biocompatibility, as the material must be non-toxic and not cause allergies, inflammation or other reactions affecting the body.

In this field, Co-Cr and Ni-Cr based alloys are often used, as they have good corrosion resistance and high wear resistance due to the crystalline nature of cobalt and nickel. In addition, Ni-Cr and Co-Cr alloys have been used in the field of dentistry for porcelain or porcelain-fused-to-metal crowns due to their good biocompatibility, wear resistance, long service life, good mechanical properties and superior corrosion.

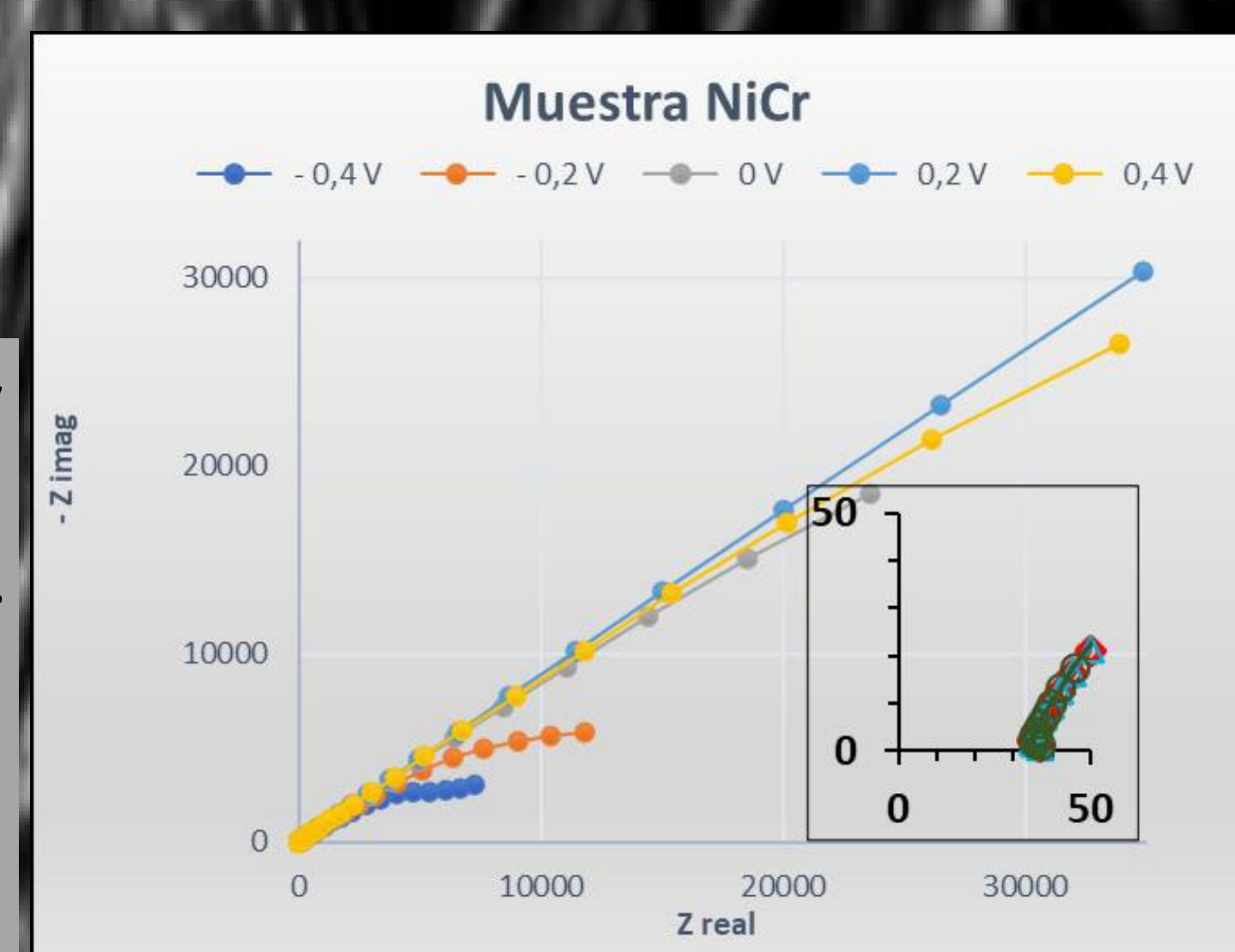
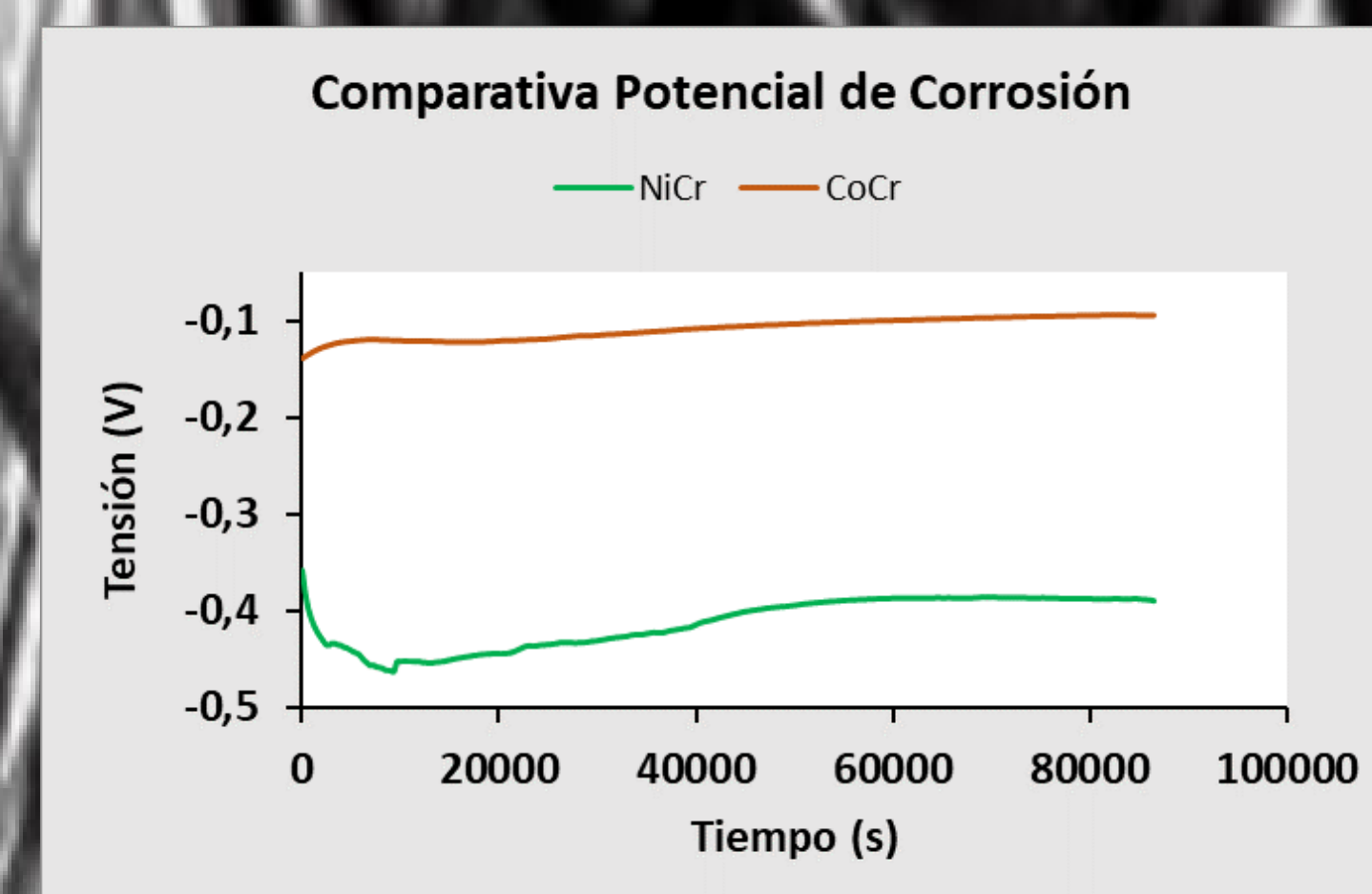
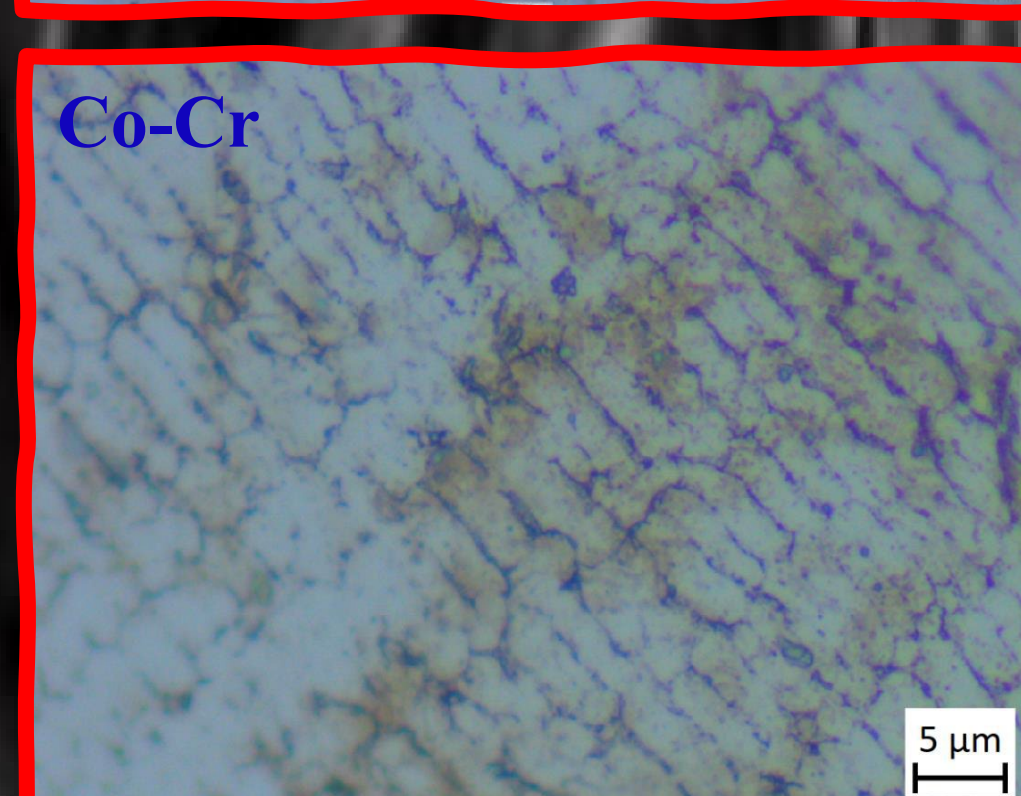
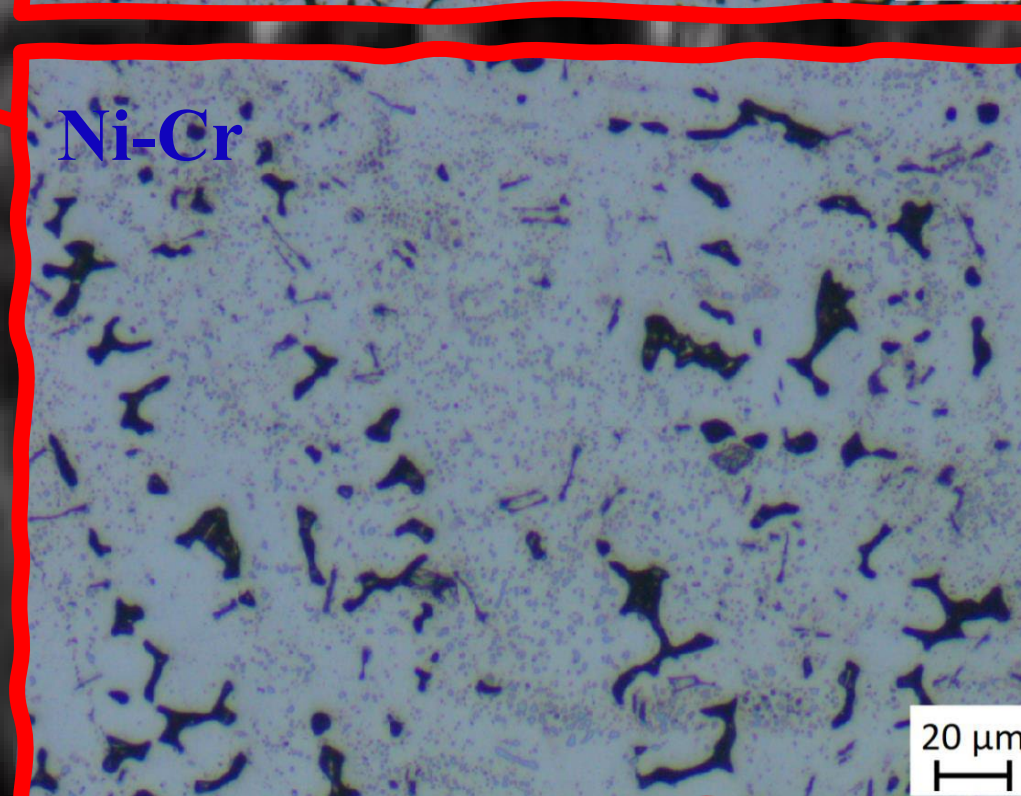
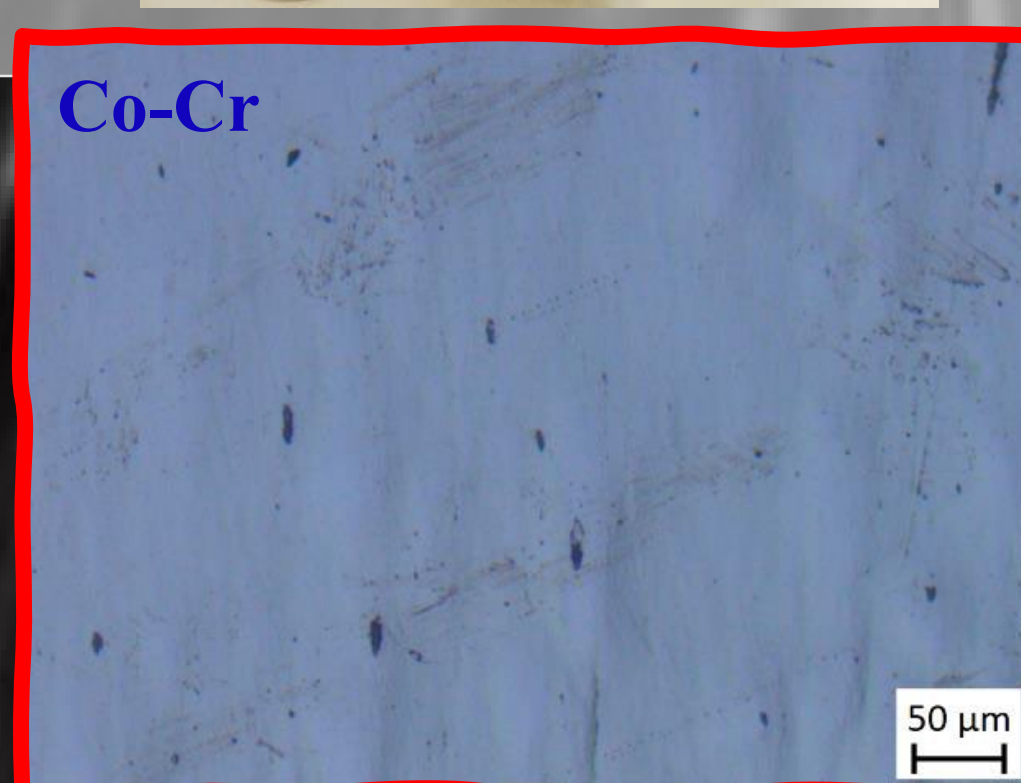
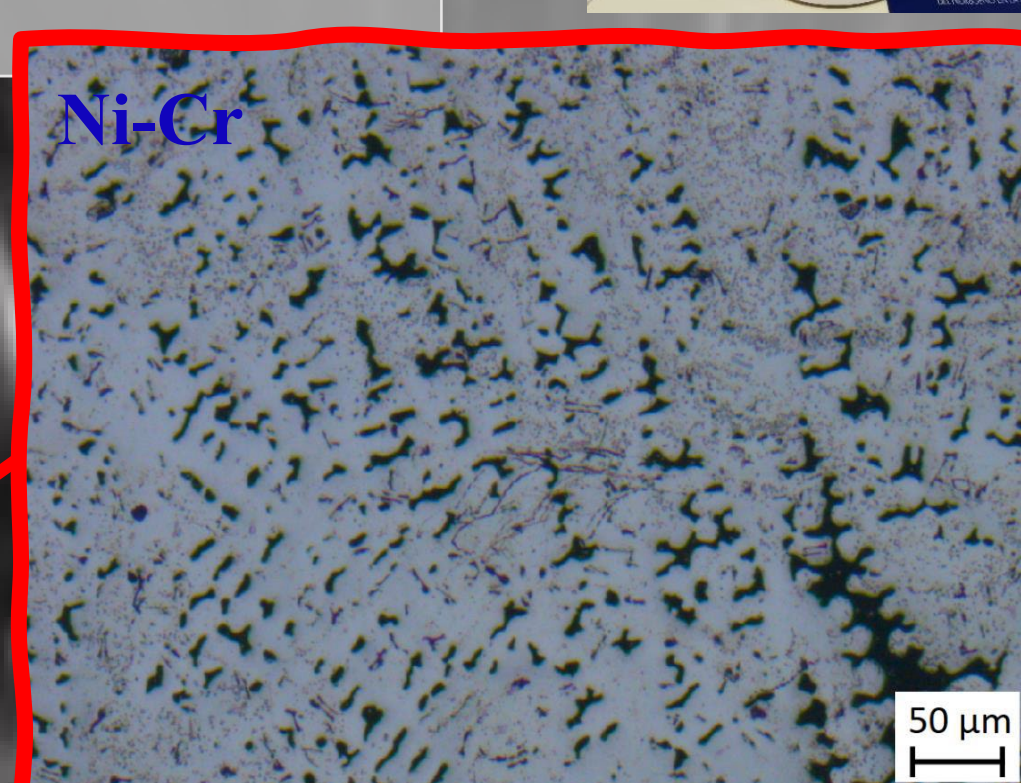
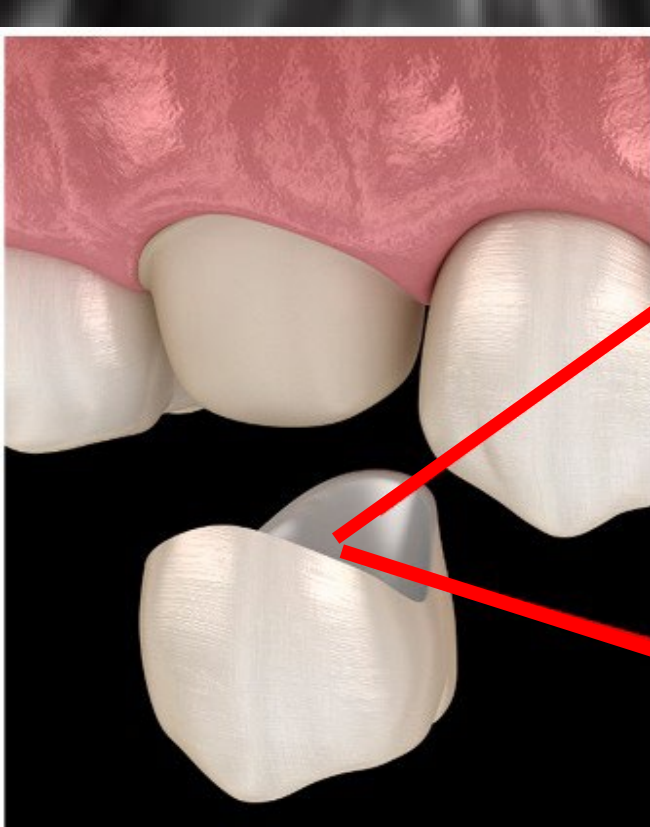
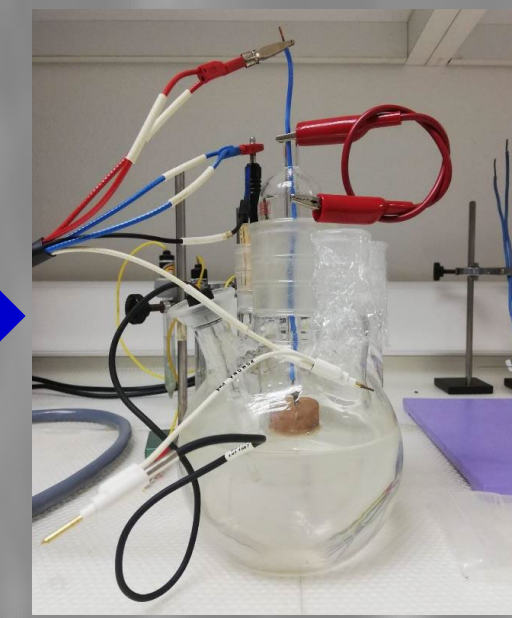
### Experimental Materials



### Surface Preparing



### Metallography Analysis



After electrochemical etching, the metallographic test is carried out by taking images with the metallographic microscope and by electrochemical test consisting in inserting a specimen into an electrochemical cell together with Saturated Calomel Electrode (SCE) as reference and Pt electrode as counter electrode. Corrosion potential ( $E_{corr}$ ) was determined and Electrochemical Impedance Spectroscopy (EIS) was applied.