

**5° COLLOQUIO INTERNAZIONALE SUL CONTROLLO
DELLA NUTRIZIONE DELLE PIANTE COLTIVATE**

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INTERACTIONSHIP CHLORIDE-NITRATE ON NUTRITION OF TOMATO PLANTS.

by

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SUMMARY.

Tomato plants var "Marglobe" were cultivated in greenhouse by hydroponic system in order to study the effect of five treatments of Cl^- , 0, 8, 16, 24 and 32 me.l^{-1} with five different concentrations of NO_3^- : 6, 12, 18, 24 and 30 me.l^{-1} . We found a cooperative effect of Cl^- and NO_3^- , increasing concentration of both until 24 me.l^{-1} of Cl^- and 18 me.l^{-1} of NO_3^- which means an increase of NO_3^- in the nutrient solution thus decreasing the toxic effect of Cl^- in plants. Org N/ NO_3^- -N in the tissues is the index that best indicates its maximum growth, its value is considered to reach 1.0.

INTRODUCTION.

The presence of Cl^- in the irrigation water is one of the problems that affect arid and semiarid areas where there is a high evapotranspiration which produces salination in the soil.

Dropping irrigation allow the control of soil solution due to slow and continuous fertilization.

Tomato plants permit a concentration of Cl^- in nutrient solution of 15 me.l^{-1} without decreasing its growth and production (Cadahia, 1968). The fertilization of tomato plants with high quantities of NO_3^- produces a decrease of Cl^- in the plant (Hernando et al. 1964). On the other hand the excess of Cl^- could produce a decrease in the NO_3^- content in the plant running the risk of inducing nitrogen deficiencies (Torres and Bingham, 1973).

Maas et al (1977) point out, using different sources, that -1.0 is the maximum osmotic potential tolerable in tomato plants.

Our work, therefore, focused in the study of Cl^- and NO_3^- relation in nutrient solutions, which allows the utilization of water with a high concentration of Cl^- by means of the increase fertilization of NO_3^- , as well as to find out to what extend Cl^- could affect assimilation, transporting and reduction of NO_3^- in tomato plants.

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MATERIAL AND METHODS.

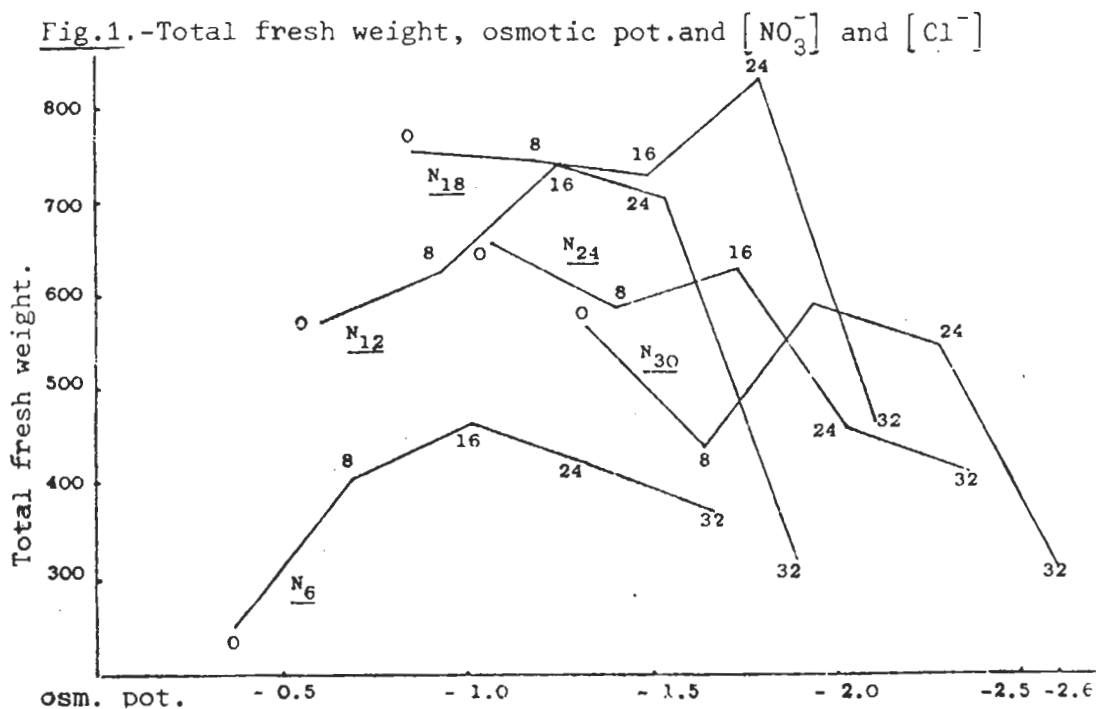
Tomato seeds (*Lycopersicum esculentum* var Marglobe) were planted in volcanic sand. After 30 days, they were transplanted in pots containing volcanic sand with an inferior draining system. Five plants per pot were planted, two were recollected at the flowering time, two at fruiting time and the fifth one was picked two weeks later in order to determine the growth.

25 different treatments were conducted in the greenhouse, using five different concentrations of Cl^- (0, 8, 16, 24, 30 me.l^{-1}). Each treatment was repeated three times. Cations were kept at the same relative level ($\text{K} : \text{Ca} : \text{Mg} = 0.35 : 0.45 : 0.20$). H_2PO_4 and $\text{SO}_4^{=}$ were kept at the same absolute level ($\text{H}_2\text{PO}_4 : 0.5 \text{ me.l}^{-1}$; $\text{SO}_4^{=} : 3.5 \text{ me.l}^{-1}$).

Cl^- , NO_3^- and organic Nitrogen were analyzed in leaves, roots and stem juice at flowering time and fruiting time. Cl^- was determined by extraction with water and valoration with silver nitrate (Ulrich and Johnson, 195). NO_3^- was extracted with water and determined by Cataldo's method modified in a higher concentration of salicylic acid in sulphuric acid (10% w/v), (Cataldo et al, 1975). Organic nitrogen was determined by microkjeldahl method.

RESULTS AND DISCUSSION.

Fig.1 depicts as an index of growth the fresh weight of the complete plant with its fruits. We found an increase of growth with the increase of NO_3^- concentration in nutrient solution until 18 me.l^{-1} , and a cooperative effect of Cl^- up to 16 me.l^{-1} of Cl^- in nutrient solution with 12 me.l^{-1} NO_3^- and 24 me.l^{-1} of Cl^- with 18 me.l^{-1} of NO_3^- . In high concentrations of NO_3^- (24 and 30 me.l^{-1}) the cooperative effect of Cl^- and NO_3^- disappears.



A equilibrate relation of Cl^- and NO_3^- increases the resistance of the plant to lower osmotic potential, achieving the best results at -1.8. Higher concentrations produce a significative decrease of the growth.

Cl^- contents in leaves, roots and stems in flowering and fruiting time (Table 1) decreases when the NO_3^- concentration in nutrient solution increases. That decrease could be produced by the interaction of Cl^- and NO_3^- passing through tonoplaste at the entry in the cell vacuoles (Cram, 1973). The decrease of Cl^- facilitates a higher resistance of the plant to the specific toxic effect of Cl^- .

The toxic concentration of Cl^- in the stem juice for tomato plants has been pointed out by Cadahia (1968) in 118 me.l^{-1} of Cl^- at flowering time and 140 me.l^{-1} of Cl^- at fruiting time and he formed these values with 15 me.l^{-1} of Cl^- in nutrient solution. In our experiment we found these concentrations in stem juice with 16 me.l^{-1} of Cl^- , 12 me.l^{-1} of NO_3^- and 24 me.l^{-1} of Cl^- - 18 me.l^{-1} of NO_3^- . This treatment ($\text{Cl}_{24}\text{N}_{18}$) is the best (Fig.1).

Values of organic Nitrogen change according to the type of the part of the plant and the sampling time. In the leaves at the flowering time there is a depressive effect of Cl^- with 6 and 12 me.l^{-1} of NO_3^- in nutrient solution. This effect is significant between $\text{Cl} 0$ and $\text{Cl} 8$ treatments. This could be possible by a decrease in the malate synthesis as a response to a high interne concentration of Cl^- , like it was pointed out by Schnabl and Raschke (1978) in stomatic cells, and that it would affect the absorption in roots and transport of NO_3^- to the shoots (Ben-Zioni et al, 1971).

In the leaves at fruiting time and in the roots at flowering time there is not a significative decrease of organic nitrogen values.

In the stem juice at flowering and fruiting time the depressive effect of Cl^- appears in higher concentrations of NO_3^- (18 , 24 , and 32 me.l^{-1}).

The relationship between org.N and $\text{NO}_3^- \text{ - N}$ (mg/mg) is the index that best agrees with the growth (Fig.2). We can see that this relationship increases with Cl^- up to 16 me.l^{-1} of Cl^- in nutrient solution with 6 and 12 me.l^{-1} of NO_3^- and 24 me.l^{-1} of Cl^- with 18 me.l^{-1} of NO_3^- .

Ferrari et al (1973) pointed out the existence of two different pools of NO_3^- in the plant cell. One of them is the vacuolar pool (metabolically inactive) and the other one is the citoplasmatic pool (metabolically active).

The competition between Cl^- and NO_3^- to enter in the vacuola may induce a decrease of total NO_3^- in the cell, but the metabolic pool can increase, in which case it would allow to an increment in the reduction of NO_3^- due to the fact that NO_3^- fluxes through the cytoplasm which regulates the activity of nitrate reductase (Shanner and Boyer, 1976).

This effect could be at the root level where we did not notice significant differences in organic nitrogen (see Table 2), due to the fact that the tomato plants may have a reduction of NO_3^- in roots (Suder-Moraw and Buczek, 1977). The reduced nitrogen could be translocated to the shoots.

Table 1

Cl⁻ contents L:leaves % d.w. R=roots % d.w. S=Stemjuice me.l.⁻¹

Cl⁻ me.l.⁻¹ in nutrient solution

		0		8		16		24		32		
		FL	FR	FL	FR	FL	FR	FL	FR	FL	FR	
NO ₃ me.l. ⁻¹ in nutrient solution	6	L	0.1	0.3	2.1	1.9	2.7	3.0	3.1	3.0	3.2	3.1
		R	0.6	0.4	2.7	3.1	3.3	3.4	4.2	4.4	3.6	3.6
		S	1.6	10	108	130	125	147	133	185	148	181
	12	L	0.2	0.3	2.1	1.8	3.0	2.9	3.0	3.3	2.9	3.2
		R	0.6	0.4	2.6	3.2	3.3	3.5	3.7	4.3	3.2	3.7
		S	1.6	17.0	100	141	118	162	122	174	118	211
	18	L	0.2	0.3	1.8	1.6	2.5	2.1	2.6	2.5	2.4	2.6
		R	0.5	0.4	2.6	2.8	2.9	3.0	3.6	3.3	3.0	4.0
		S	1.6	21	81	123	93.2	127	116	139	101	161
	24	L	0.2	0.3	1.4	1.5	1.7	1.7	1.8	2.3	2.1	2.7
		R	0.5	0.3	2.3	2.1	2.5	2.6	2.5	4.3	3.0	3.8
		S	1.6	19	66	77	76	126	89	171	91	178
	30	L	0.2	0.3	1.0	1.4	1.4	1.5	1.7	2.6	2.0	2.6
		R	0.5	0.4	1.7	2.2	1.8	2.1	2.7	4.0	3.1	3.9
		S	1.6	17	55	96	74	92	87	171	90.0	197

FL= Flowering time FR= Fruiting time

Table 2

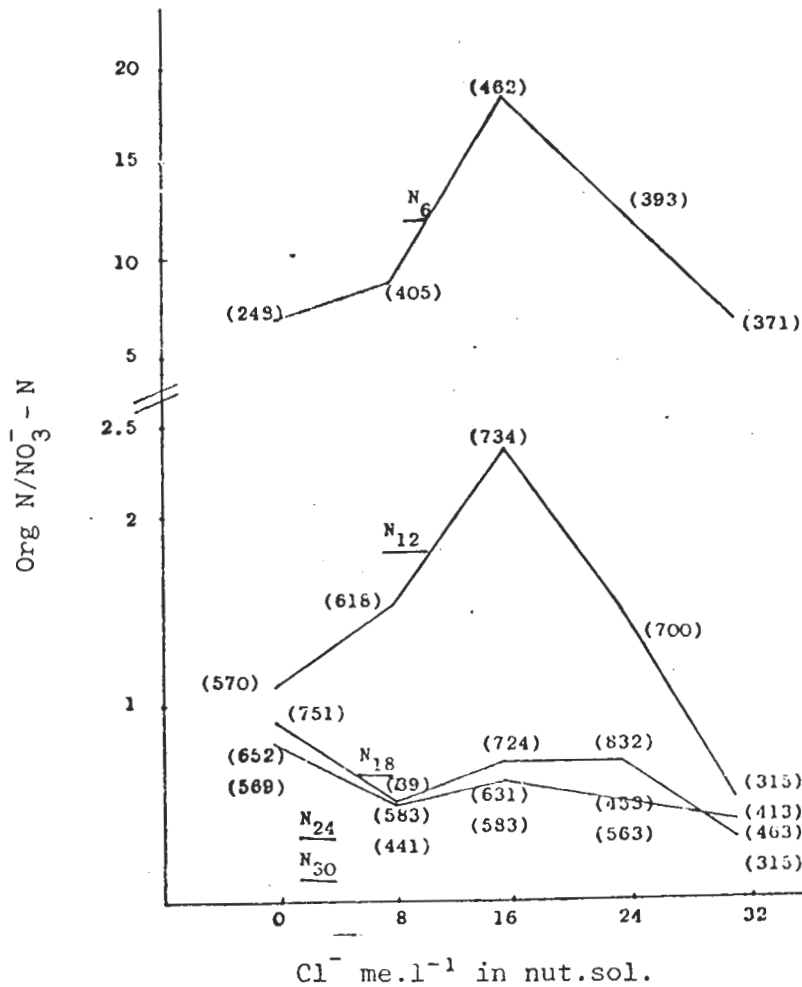
Org N. L=leaves % in d.w. R=roots % in d.w. S=Stemjuice g.l⁻¹

Cl⁻ me.l.⁻¹ in nutrient solution

		0		8		16		24		32		
		FL	FR	FL	FR	FL	FR	FL	FR	FL	FR	
NO ₃ (me.l ⁻¹) in nutrient solution	6	L	5.5	4.0	3.5	3.4	3.3	3.4	3.4	3.5	3.2	3.3
		R	1.7	1.8	1.6	1.5	1.7	1.5	1.8	1.5	1.8	1.6
		S	0.4	0.7	0.3	0.3	0.3	0.3	0.3	0.3	0.3	0.3
	12	L	5.5	4.3	4.4	4.2	3.7	4.3	3.9	4.1	4.2	4.2
		R	1.7	2.2	1.9	1.8	1.7	1.8	2.0	1.9	2.3	1.8
		S	0.5	0.8	0.4	0.5	0.5	0.4	0.4	0.5	0.4	0.4
	18	L	5.1	4.6	4.5	4.4	4.1	4.4	4.5	4.4	4.3	3.8
		R	2.3	2.3	2.1	1.9	2.4	1.9	2.1	2.0	2.3	1.9
		S	0.8	0.8	0.5	0.5	0.5	0.4	0.6	0.8	0.4	0.4
	24	L	4.5	4.5	4.3	4.7	4.4	5.2	4.8	4.6	5.0	4.6
		R	2.4	2.2	2.0	2.0	1.9	2.2	2.1	2.2	2.3	2.4
		S	0.7	0.8	0.5	0.5	0.6	0.4	0.6	0.4	0.6	0.6
	30	L	5.2	4.7	4.1	3.8	4.5	4.7	5.0	3.9	4.6	4.1
		R	2.3	2.7	2.3	2.6	2.4	2.3	2.6	2.1	2.2	2.3
		S	1.0	0.8	0.5	0.3	0.7	0.4	0.6	0.5	0.7	0.7

FL= Flowering time FR= Fruiting time

Fig.2.- Variation of the relation Organic N/NO₃⁻-N in stem juice at the flowering time (numbers in brackets are total fresh weights of each treatment).



CONCLUSIONS.

The increment of Cl⁻ in the culture medium could be partially mitigated by the increase of NO₃⁻ fertilization in the tomato culture.

In addition this enhances a better resistance to the effect of total salinity.

The cooperative effect of Cl⁻ and NO₃⁻ could be explained by an increase of reduction in the roots and the translocation to the shoots of the reduced nitrogen.

The relation Org N/NO₃⁻-N in stem juice at flowering time is a good index to study the effect of Cl⁻ in the NO₃⁻ metabolism and its value should be maintained next to 1.0. However, we believe that these results ought to be further studied in order to yield applicable results.

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RÉSUMÉ : INTERACTION DU CLORURE ET NITRATE EN TOMATES.

Des tomates de la variété Marglobe ont été cultivées avec une solution nutritive dans une serre pour l'étude de l'effet du Cl^- . Cinq traitements avec du Cl^- : 0,8,16,24 et 32 me.l^{-1} ont été combinés avec cinq concentrations différentes de NO_3^- : 6,12,18,24 et 30 me.l^{-1} . On trouve un effet coopératif du Cl^- et du NO_3^- en augmentant les deux concentrations, celle du Cl^- jusqu'à 24 me.l^{-1} et celle du NO_3^- jusqu'à 18 me.l^{-1} . Ce qui indique qu'en augmentant le NO_3^- dans la solution l'effet toxique du Cl^- diminue. L'index qui s'ajuste le mieux aux rendements c'est celui de la relation $\text{N.org} : \text{N-NO}_3^-$ (mg:mg) dont les valeurs doivent être proches à 1.

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